

## CHEMISTRY UNIT 1 STUDY GUIDE.

### What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

### Types

- Substance- a particular kind of matter – pure- made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

### Properties:

**Physical** – property that can be observed and measure without changing the substances.

**Chemical** – property that can only be observed by changing the type of substance.

**Extensive**- only depends on the amount of matter

Mass, weight, volume, length

**Intensive** – only depends on the type of matter, not the amount used to identify a substance.

Color, Oder, melting/ freezing point.

**Ductility**- the ability to make copper into wires.

### States of Matter:

	Definite volume	Definite shape	Temperature increase	compressable
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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School**

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*Courses* Institute for European Environmental Policy,Commission of the European Communities. Directorate-General Environment, Nuclear Safety and Civil Protection,1993      **The McGuffey Reader** ,1937      *Science Education in Canadian*

*Schools. Volume III. Case Studies of Science Teaching. Background Study* 52 John Olson,Science Council of Canada, Ottawa (Ontario).,1984 Volume 3 of a three volume study of science education in Canadian schools reporting 8 case studies of

science teaching in action in Canadian schools Sites for intensive study were selected to include a diversity of both regions and school settings and at each school site various kinds of information were collected to obtain as complete a picture as

possible Major issues addressed include integration and options as forms of curriculum organization socialization as a priority the inquiry approach and understanding and the dynamics of change and dilemmas of practice      *Chemistry and*

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