

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and pOH and vice versa.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                   |   |                  |
|-------------------|---|------------------|
| 1. autoionization | 2. hydronium ion ( $\text{H}_3\text{O}^+$ ) | 3. acid solution |
| 4. autoionization | 5. hydronium ion ( $\text{H}_3\text{O}^+$ ) | 6. acid solution |
|                   |   |                  |

## Key Equation

- |                                    |  |
|------------------------------------|--|
| 1. $\text{pH} = -\log[\text{H}^+]$ | 2. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ |
| 3. $\text{pH} = -\log[\text{H}^+]$ | 4. $\text{pH} = -\log[\text{H}_3\text{O}^+]$ |

## Part A Completion

Write the equation and/or the numerical relationship of the values indicated that are indicated in the space provided. Be sure to include units and the appropriate symbols.

1. What substance is  $\text{H}^+$  in water? Hydrogen ion.
2. What is the pH of a 0.1 M solution of HCl? 1.
3. What is the pH of a 0.1 M solution of NaOH? 13.
4. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
5. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
6. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
7. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
8. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
9. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
10. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
11. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
12. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
13. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
14. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
15. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
16. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
17. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
18. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.
19. What is the pH of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$ ? 11.
20. What is the pH of a 0.1 M solution of  $\text{H}_2\text{SO}_4$ ? 1.

1.  $\text{H}^+$
- 1
- 13
- 1
- 11
- 1
- 11
- 1
- 11
- 1
- 11
- 1
- 11
- 1
- 11
- 1
- 11
- 1
- 11
- 1

# 19 2 Hydrogen Ions And Acidity Answer Key

**Daniel F McAuley**



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