

6. Which of the following structures is the same as structure A?



A



A



B



C



D

7. Which of the following statements best describes the carbon-chlorine bond of the molecule below?



A. nonpolar; no dipole
B. polar; δ+ at carbon and δ- at chlorine
C. polar; δ- at carbon and δ+ at chlorine
D. ionic
E. none of the above

8. The lone pairs of electrons of the oxygen atom are located in which orbitals?



A. sp^2
B. sp^3
C. sp
D. s
E. p

9. Which statement concerning the length of bonds indicated in the molecule below is correct?



A. II is the shortest and I is the longest
B. II is the shortest and III is the longest
C. III is the shortest and II is the longest
D. III is the shortest and I is the longest
E. I is the shortest and III is the longest

10. Select the series of atoms that is correctly arranged in order of increasing electronegativity.

A. C < N < B < Br
B. P < N < As < F
C. Li < B < N < F

D. Cl < Cs < C < Co
E. Be < B < Ba < Br

11. What is the geometry of nitrogen in the following molecule?



A. tetrahedral
B. trigonal planar
C. trigonal pyramidal
D. bent
E. linear

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**Professor Nicole Graulich,Dr Ginger
Shultz**

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overview of the issues facing new chemistry faculty in preparation for teaching. Serving as a reference to answer specific questions new chemistry faculty encounter this book is comparable to sitting down with a colleague in the department and talking through some ideas or gaining some pointers on how to avoid common pitfalls. It is the one single place new chemistry faculty can go to find practical information on how to teach and how to prepare for teaching their first course. Chapters are written both by established experts in the field and by new professors within their first couple of years of teaching. **ACS Organic Chemistry Exams - The Official Guide** Examinations Institute, 2002

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