

6. Which of the following structures is the same as structure A?



A



A



B



C



D

7. Which of the following statements best describes the carbon-chlorine bond of the molecule below?



- A. nonpolar; no dipole
 B. polar; $\delta+$ at carbon and $\delta-$ at chlorine
 C. polar; $\delta-$ at carbon and $\delta+$ at chlorine
 D. ionic
 E. none of the above

8. The lone pairs of electrons of the oxygen atom are located in which orbitals?



- A. sp^2
 B. sp^3
 C. sp
 D. s
 E. p

9. Which statement concerning the length of bonds indicated in the molecule below is correct?



- A. II is the shortest and I is the longest
 B. II is the shortest and III is the longest
 C. III is the shortest and II is the longest
 D. III is the shortest and I is the longest
 E. I is the shortest and III is the longest

10. Select the series of atoms that is correctly arranged in order of increasing electronegativity.

- A. $C < N < B < Br$
 B. $P < N < As < F$
 C. $Li < B < N < F$
 D. $Cl < Cs < C < Co$
 E. $Be < B < Ba < Br$

11. What is the geometry of nitrogen in the following molecule?



- A. tetrahedral
 B. trigonal planar
 C. trigonal pyramidal
 D. bent
 E. linear

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**Professor Nicole Graulich, Dr Ginger
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