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ppm: Answer- (grams analyte/grams sample) $\times 10^6$

Molarity: Answer- moles analyte/liter of solution

Volume Percent: Answer- (volume solute/volume solution) $\times 100$

Volume ppm: Answer- (volume solute/volume solution) $\times 10^6$

kilo-: Answer- 10^3

deci-: Answer- 10^{-1}

centi-: Answer- 10^{-2}

milli-: Answer- 10^{-3}

micro-: Answer- 10^{-6}

nano-: Answer- 10^{-9}

pico-: Answer- 10^{-12}

femto-: Answer- 10^{-15}

weight percent: Answer- (grams analyte/grams sample) $\times 100$

ppt: Answer- (grams analyte/grams sample) $\times 10^3$

ppt simplified: Answer- gram analyte/liter solution

Acs Elementary Analytical Chemistry Exam Study Guide

Clifford Lane

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