

ACS Final Exam Practice 1

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is  $3.0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$ . What is this speed in  $\text{nm}\cdot\text{ms}^{-1}$ ?

- (A)  $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$       (B)  $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$       (C)  $3.0 \times 10^{14} \text{ nm}\cdot\text{ms}^{-1}$       (D)  $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

- (A) 1.3 g      (B) 1.30 g      (C) 1.298 g      (D) 1.2980 g

PQ-5. What is the name of  $\text{Ti}_2(\text{PO}_4)_3$ ?

- (A) titanium phosphate      (B) titanium(III) phosphate  
(C) titanium(IV) phosphate      (D) titanium tetrophosphate

PQ-6. The formula of strontium hexafluorosilicate is  $\text{SrSiF}_6$ . What is the formula of aluminum hexafluorosilicate?

- (A)  $\text{AlSiF}_6$       (B)  $\text{Al}_3(\text{SiF}_6)_3$       (C)  $\text{Al}_2\text{SiF}_6$       (D)  $\text{Al}_3(\text{SiF}_6)_2$

PQ-8. What is the correct name for  $\text{N}_2\text{O}_5$ ?

- (A) nitrogen(II) oxide      (B) nitrogen(V) oxide  
(C) dinitrogen oxide      (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of  $12.0 \text{ g}\cdot\text{cm}^{-3}$ , what is the mass of a nanoparticle with a volume of  $1.84 \times 10^{-23} \text{ cm}^3$ ?

- (A)  $2.21 \times 10^{-23} \text{ g}$       (B)  $1.63 \times 10^{-20} \text{ g}$       (C)  $6.52 \times 10^{-21} \text{ g}$       (D)  $1.53 \times 10^{-22} \text{ g}$

# Acs Chemistry 2 Answer Key

**Linda L. Murphy**



## Acs Chemistry 2 Answer Key:

*TEXT BOOK OF MEDICINAL CHEMISTRY-II* Mr. Sudam Gorakhnath Mule, Mr. Jagdish Atmaram Waghmare, 2024-11-06

The textbook Medicinal Chemistry II offers a detailed exploration of various therapeutic agents and their pharmacological actions. It begins with antihistaminic agents discussing histamine its receptors and a variety of H1 and H2 antagonists. It covers proton pump inhibitors and extends into antineoplastic agents with a focus on alkylating agents antimetabolites and plant derived products. Anti anginal drugs including vasodilators and calcium channel blockers are examined next followed by a section on diuretics. Other topics include antihypertensive drugs antiarrhythmics antihyperlipidemic agents coagulants anticoagulants and drugs for treating congestive heart failure. The book also delves into endocrine system agents with sections on sex hormones corticosteroids and drugs for erectile dysfunction. Antidiabetic agents including insulin and oral hypoglycemics are thoroughly covered. The text concludes with local anesthetics emphasizing their structure activity relationships and various chemical classes.

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Comprehensive Supramolecular Chemistry II George W. Gokel, Len Barbour, 2017-06-22. Comprehensive Supramolecular Chemistry II Second Edition Nine Volume Set is a one stop shop that covers supramolecular chemistry a field that originated from the work of researchers in organic inorganic and physical chemistry with some biological influence. The original edition was structured to reflect in part the origin of the field. However in the past two decades the field has changed a great deal as reflected in this new work that covers the general principles of supramolecular chemistry and molecular recognition experimental and computational methods in supramolecular chemistry supramolecular receptors dynamic supramolecular chemistry supramolecular engineering crystallographic engineered assemblies sensors imaging agents devices and the latest in nanotechnology. Each section begins with an introduction by an expert in the field who offers an initial perspective on the development of the field. Each article begins with outlining basic concepts before moving on to more advanced material. Contains content that begins with the basics before moving on to more complex concepts making it suitable for advanced undergraduates as well as academic researchers. Focuses on application of the theory in practice with particular focus on areas that have gained increasing importance in the 21st century including nanomedicine nanotechnology and medicinal chemistry. Fully rewritten to make a completely up to date reference work that covers all the major advances that have taken place since the First Edition published in 1996.

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**Symposium of Materials Science and Chemistry II** Tutik Dwi Wahyuningsih, Roto Roto, Rohana Adnan, Laurent Commeiras, Kuwat Triyana, Indriana Kartini, Julius Motuzas, Dwi Siswanta, 2020-04-24. 5th International Conference on Science and Technology ICST 2019. Selected peer reviewed papers from the 5th International Conference on Science and Technology ICST 2019 July 30-31 2019 Yogyakarta Indonesia.

*Essentials of Physical Chemistry* Don Shillady, 2011-07-27 At a time when U S high school students are producing low scores in mathematics and science on international examinations a thorough grounding in physical chemistry should not be considered optional for science undergraduates Based on the author s thirty years of teaching *Essentials of Physical Chemistry* merges coverage of calculus with chemistry and molecular physics in a friendly yet thorough manner Reflecting the latest ACS guidelines the book can be used as a one or two semester course and includes special topics suitable for senior projects The book begins with a math and physics review to ensure all students start on the same level and then discusses the basics of thermodynamics and kinetics with mathematics tuned to a level that stretches students abilities It then provides material for an optional second semester course that shows students how to apply their enhanced mathematical skills in a brief historical development of the quantum mechanics of molecules Emphasizing spectroscopy the text is built on a foundation of quantum chemistry and more mathematical detail and examples It contains sample classroom tested exams to gauge how well students know how to use relevant formulas and to display successful understanding of key concepts Coupling the development of mathematical skills with chemistry concepts encourages students to learn mathematical derivations Mini biographies of famous scientists make the presentation more interesting from a people point of view Stating the basic concepts of quantum chemistry in terms of analogies provides a pedagogically useful technique Covering key topics such as the critical point of a van der Waals gas the Michaelis Menten equation and the entropy of mixing this classroom tested text highlights applications across the range of chemistry forensic science pre medical science and chemical engineering In a presentation of fundamental topics held together by clearly established mathematical models the book supplies a quantitative discussion of the merged science of physical chemistry Tests in Print Oscar Krisen Buros, 2006 **Science Tests and Reviews** Buros Center, 1975 Science Tests and Reviews consisting of science sections of the first seven MMYs and Tests in Print II includes 217 original test reviews written by 81 specialists 18 excerpted test reviews 270 references on the construction use and validity of specific tests a bibliography on in print science tests references for specific tests cumulative name indexes for specific tests with references a publishers directory title index name index and a scanning index The 97 tests covered fall into the following categories 23 general 14 biology 35 chemistry 3 geology 6 miscellaneous and 16 physics *Tests in Print V* Linda L. Murphy, James C. Impara, Barbara S. Plake, 1999 Customers who place a standing order for the Tests in Print series or the Mental Measurements Yearbook series will receive a 10% discount on every volume To place your standing order please call 1 800 848 6224 in the U S or 919 966 7449 outside the U S Designed to complement the Mental Measurements Yearbooks Tests in Print fills a pressing need for a comprehensive bibliography of all commercially available English language tests in print Although these volumes are useful in and of themselves their maximum usefulness requires the availability and use of the Mental Measurements Yearbooks Although information on available tests and specific test bibliographies is valuable the greatest service which Tests in Print

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