

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

1. What is a heterogeneous mixture? Provide two examples.

2. Explain the characteristics of a saturated solution and an unsaturated solution.

3. Define the term solvent and provide an example.

4. What is solubility, and how is it measured?

5. Distinguish between concentrated and dilute.

6. What are the properties of acids?

7. What are the properties of bases?

8. What is the pH scale?

9. What are the characteristics of acids?

10. What are the characteristics of bases?

11. What is the pH of a neutral solution?

12. List the properties of bases and provide an example.

13. Explain the significance of the pH scale.

14. How does the concentration of an acid affect its pH?

15. How does the concentration of a base affect its pH?

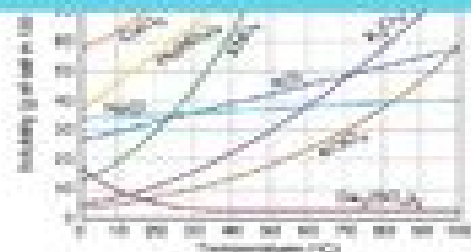
16. What is the pH of a strong acid?

17. What is the pH of a strong base?

18. What is the pH of a weak acid?

19. What is the pH of a weak base?

20. What is the pH of a neutral solution?



21. Describe how you would use the graph to determine the amount of Na_2SO_4 that can be dissolved in 100 grams of water at 40°C .

22. Which salt has the highest solubility at 100°C ?



ANSWER KEY INCLUDED

Acids And Bases Chemistry Guide Key

PT Brinkman



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