

Multiple Choice (20 questions, 4% each): Mark all answers on your Scantron form.

1. Which of the following is NOT TRUE of an elemental substance?

- (a) It must have an element symbol that can be found on the Periodic Table ✓
- (b) It must be composed of a single type of atom ✓
- (c) It cannot be separated into simpler substances by chemical means ✓
- (d) It must be a molecular substance. *- Not elements are mostly not molecules*
- (e) It can be defined by a specific atomic number ✓

2. Hydrogen reacts with oxygen to form water. Water is a compound (defined later) of hydrogen and oxygen, and the formation of water from hydrogen and oxygen involves a change. *- reactions are chemical changes*

- (a) Compound, physical
- (b) Mixture, physical
- (c) Compound, chemical
- (d) Mixture, chemical
- (e) Elemental substance, physical

3. Which of the following quantities is correctly expressed to four significant figures?

- (a) 0.3679×10^{-2} mmHg *- not good the leading zero is not leading zero*
- (b) 9,437 g Q.E. *4 sf*
- (c) 0.002 cm *2 sf*
- (d) 2.30012×10^5 Pa *6 sf*
- (e) 0722 m *3 sf*

4. How many mL of water are in a sample that has a volume of 22.4 L?

- (a) 0.0224 mL
- (b) 0.224 mL
- (c) 22.4 mL
- (d) 2240 mL
- (e) 2.24×10^4 mL

$$\frac{22.4 \text{ L}}{1 \text{ L}} \times \frac{1000 \text{ mL}}{1 \text{ L}} = 22,400 = 2.24 \times 10^4 \text{ mL}$$

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