

# Acids, Bases, Salts

1. The acid ion	A. 10
2. The base ion	B. 7
3. pH value of an acid	C. titration
4. pH value of a base	D. NaCl
5. Neutral pH value	E. Neutralization
6. Acid taste	F. 0-<7
7. Base taste	G. H <sup>+</sup>
8. Color of phenolphthalein in an acid	H. >7-14
9. Color of phenolphthalein in a base	I. Base
10. Technique used to find the unknown concentration of an acid solution	J. Red
11. Feels slippery	K. 6
12. Reacts with metals to produce hydrogen gas	L. Bitter
13. Formed from neutralization	M. Sour
14. Reaction when the number of H <sup>+</sup> = OH <sup>-</sup>	N. Salt and Water
15. An example of an acid	O. Pink
16. An example of a base	P. Table K
17. An example of a salt	Q. Acid
18. Color of litmus in an acid	R. Blue
19. Color of litmus in a base	S. Colorless
20. Reference table of acids	T. NaOH
21. pH that is 100X more acidic than a pH of 8	U. HBr
22. pH that is 100X more basic than a pH of 8	V. OH <sup>-</sup>

# Acids Bases And Salts Review Sheet

**Leo J. Malone, Theodore O. Dolter**

## Acids Bases And Salts Review Sheet:

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