

Study Guide Unit 9

1. KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):

- a. Solution- m. Solute-
- b. Homogeneous- n. Solvent-
- c. Heterogeneous- o. Acid-
- d. Soluble- p. Base-
- e. Insoluble- q. Titration-
- f. Miscible- r. Analyte-
- g. Immiscible- s. Titrant-
- h. Solubility- t. Monoprotic acid-
- i. Alloy- u. Diprotic Acid
- j. Saturated- v. Triprotic Acid
- k. Unsaturated-
- l. Supersaturated-

2. How is the solubility of a gas affected by temp. and pressure?

Pressure up = solubility up

Temp up = solubility down

3. Under what conditions are salts most soluble?

High temperatures

4. Use the solubility curve to the right to answer the following.

a. What mass of KNO_3 will make a saturated solution at 60°C ? **100g**

b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases?

NH_3 , $\text{Ca}_2(\text{SO}_4)_3$

c. What type of solution do I have if I **add** 110 g of NaNO_3 in 100g of 20°C water?

Saturated

d. What mass of KCl can go into solution at 80°C ?

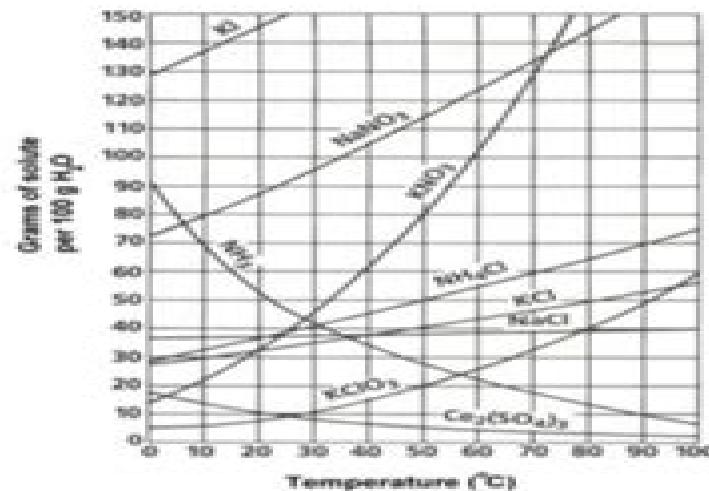
50 g

e. What is the least soluble compound at 20°C ?

KClO_3

f. Excluding KI , what is the most soluble compound at 80°C ?

KNO_3



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NARAYAN CHANDER

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