

Assessment

Acids and Bases

Section Quiz: Acid-Base Theories

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. All Bronsted- Lowry acids
 - a. are aqueous solutions.
 - b. can act as Arrhenius acids.
 - c. donate protons.
 - d. All of the above
- _____ 2. Which of the following substances is both a Bronsted-Lowry base and an Arrhenius base?
 - a. $\text{NH}_3(s)$
 - b. $\text{NH}_3(aq)$
 - c. $\text{HCl}(g)$
 - d. $\text{HCl}(aq)$
- _____ 3. In the following reaction, which substance acts as a Bronsted-Lowry acid?

$$\text{Ba(OH)}_2(s) + \text{H}_2\text{SO}_4(s) \rightarrow \text{H}_2\text{O}(l) + \text{BaSO}_4(s)$$

Acid Base Theories Section Review Answers

John McMurry

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