

Assessment

# Acids and Bases

## Section Quiz: Acid-Base Theories

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. All Brønsted- Lowry acids
  - a. are aqueous solutions.
  - b. can act as Arrhenius acids.
  - c. donate protons.
  - d. All of the above
- \_\_\_\_\_ 2. Which of the following substances is both a Brønsted-Lowry base and an Arrhenius base?
  - a.  $\text{NH}_3(s)$
  - b.  $\text{NH}_3(aq)$
  - c.  $\text{HCl}(g)$
  - d.  $\text{HCl}(aq)$
- \_\_\_\_\_ 3. In the following reaction, which substance acts as a Brønsted-Lowry acid?



- a.  $\text{HClO}_4$
- b.  $\text{H}_2\text{O}$
- c.  $\text{H}_3\text{O}^+$
- d.  $\text{ClO}_4^-$

- \_\_\_\_\_ 4. In the following reaction, which substance acts as a Brønsted-Lowry base?



- a.  $\text{HClO}_4$
- b.  $\text{H}_2\text{O}$
- c.  $\text{H}_3\text{O}^+$
- d.  $\text{ClO}_4^-$

- \_\_\_\_\_ 5. Which of the following is a polyprotic acid?

- a.  $\text{HCl}$
- b.  $\text{H}_2\text{SO}_4$
- c.  $\text{HNO}_3$
- d.  $\text{H}_2\text{O}$

# Acid Base Theories Section Review Answers

**John McMurry**



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## **Table of Contents Acid Base Theories Section Review Answers**

1. Understanding the eBook Acid Base Theories Section Review Answers
  - The Rise of Digital Reading Acid Base Theories Section Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Acid Base Theories Section Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acid Base Theories Section Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acid Base Theories Section Review Answers
  - Personalized Recommendations
  - Acid Base Theories Section Review Answers User Reviews and Ratings
  - Acid Base Theories Section Review Answers and Bestseller Lists
5. Accessing Acid Base Theories Section Review Answers Free and Paid eBooks
  - Acid Base Theories Section Review Answers Public Domain eBooks
  - Acid Base Theories Section Review Answers eBook Subscription Services
  - Acid Base Theories Section Review Answers Budget-Friendly Options
6. Navigating Acid Base Theories Section Review Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Acid Base Theories Section Review Answers Compatibility with Devices
  - Acid Base Theories Section Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Acid Base Theories Section Review Answers
  - Highlighting and Note-Taking Acid Base Theories Section Review Answers
  - Interactive Elements Acid Base Theories Section Review Answers

8. Staying Engaged with Acid Base Theories Section Review Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Acid Base Theories Section Review Answers
9. Balancing eBooks and Physical Books Acid Base Theories Section Review Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Acid Base Theories Section Review Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Acid Base Theories Section Review Answers
  - Setting Reading Goals Acid Base Theories Section Review Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acid Base Theories Section Review Answers
  - Fact-Checking eBook Content of Acid Base Theories Section Review Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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