

Observations & Calculations:**Part A: Calibration of pipette:**

1. Laboratory temperature = _____ °C
2. Density of water at the laboratory temperature = _____ gcm⁻³.

Marked capacity of the pipette	Weight of beaker	Weight of beaker + water	Weight of water	Observed volume or True capacity of the pipette V ₁ cm ³	Correction
V ₁ cm ³	w ₁	w ₂	(w ₂ - w ₁)	$\frac{(w_2 - w_1)}{\text{Density of water}}$	(V ₁ ± V ₂) cm ³
25 cm ³					

Part B: Calibration of burette:

1. Laboratory temperature = _____ °C
2. Density of water at the laboratory temperature = _____ gcm⁻³.

Trial No.	Burette readings		Volume of water [Marked capacity] in cm ³	Mass of beaker	Mass of beaker + water	Mass of water	Observed volume or True capacity V ₂ cm ³	Correction
	Final	Initial						
			V ₁	w ₁	w ₂	(w ₂ - w ₁)	$\frac{(w_2 - w_1)}{\text{Density of water}}$	(V ₁ ± V ₂) cm ³
1	10.0	00.0	10.0					
2	20.0	10.0	10.0					
3	30.0	20.0	10.0					
4	40.0	30.0	10.0					
5	50.0	40.0	10.0					

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