

To Build a Fire

by Jack London

Day had broken cold and grey, exceedingly cold and grey, from the main Yukon trail and climbed the high earth-bar travelled trail led eastward through the fat spruce timber! he paused for breath at the top, excusing the act to himself: it was nine o'clock. There was no sun nor hint of sun, though sky, it was a clear day, and yet there seemed an intangible subtle gloom that made the day dark, and that was due to the fact that the sun did not worry the man. He was used to the lack of sun. It had not been the sun, and he knew that a few more days must pass, would just peep above the sky-line and dip immed-

The man flung a look back along the way he had come, hidden under three feet of ice. On top of this ice were a pure white, rolling in gentle undulations where the ice-jar North and south, as far as his eye could see, it was unlike the dark and twisted from around the spruce-ct that curved and twisted away into the north, where I spruce-covered island. This dark hairline was the trail—five hundred miles to the Chilkoot Pass, Dyea, and salt-w miles to Dawson, and still on to the north a thousand miles to Michael on Bering Sea, a thousand miles and half a thou-

But all this—the mysterious, far-reaching hairline trail, the tremendous cold, and the strangeness and weirdness impression on the man, it was not because he was long used to it. He was a new-comer in the land, a chechague, and this was his first

The Tell-Tale Heart

by Edgar Allan Poe

True!—nervous—very, very dreadfully nervous I had been and am; but why will you say that I am mad? The disease had sharpened my senses—not destroyed—not dulled them. Above all was the sense of hearing acute. I heard all things in the heaven and in the earth. I heard many things in hell. How, then, am I mad? Hearken! and observe how healthily—how calmly I can tell you the whole story.

It is impossible to say how first the idea entered my brain; but once conceived, it haunted me day and night. Object there was none. Passion there was none. I loved the old man. He had never wronged me. He had never given me insult, for his gold I had no desire. I think it was his eye! yes, it was this! He had the eye of a vulture—a pale blue eye, with a film over it. Whenever it fell upon me, my blood ran cold; and so, by degrees—very gradually—I made up my mind to take the life of the old man, and thus rid myself of the eye forever.

Now this is the point, You fancy me mad. Madmen know nothing. But you should have seen me. You should have seen how wisely I proceeded—with what caution—with what foresight—with what dissimulation I went to work! I was never kinder to the old man than during the whole week before I killed him. And every night, about midnight, I turned the latch of his door and opened it—oh, so gently! And then, when I had made an opening sufficient for my head, I put in a dark lantern, all closed, closed, that no light shone out, and then I thrust in my head. Oh, you would have laughed to see how cunningly I thrust it in! I removed it slowly—very, very slowly, so that I might not disturb the old



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calorimeter with a total heat capacity of 15 48 kJ C the temperature of the calorimeter and its contents changes from 19 77 C to 33 90 C calculate e and h for the following reaction as written 2 NH₃ (g) + 3 O₂ (g) 2 NO₂ (g)

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web choose 1 answer the final temperature is closer to t₁ than to t₂ a the final temperature is closer to t₁ than to t₂ the final temperature is exactly halfway between t₁ and t₂ b the final temperature is exactly halfway between t₁ and t₂ the final temperature is closer to t₂ than to t₁ c

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web in this set of practice questions we will go over the main types of questions on calorimetry including the heat capacity the heat of reaction finding the final temperature of a mixture constant pressure calorimetry and constant volume calorimetry a few important concepts and formulas you will need to solve these questions

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web 6 a 218 g sample of steam at 121.0 C is cooled to ice at 14.0 C find the change in heat content of the system 7 in going from ice at 34.0 C to steam at 138.0 C a sample of water absorbs 1.41 x 105 J find the mass of the sample answers 1 2 68 x 104 J 2 6 23 x 104 J 3 1 71 x 104 J 4 1 11 x 105 J 5 2 11 x 105 J 6

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