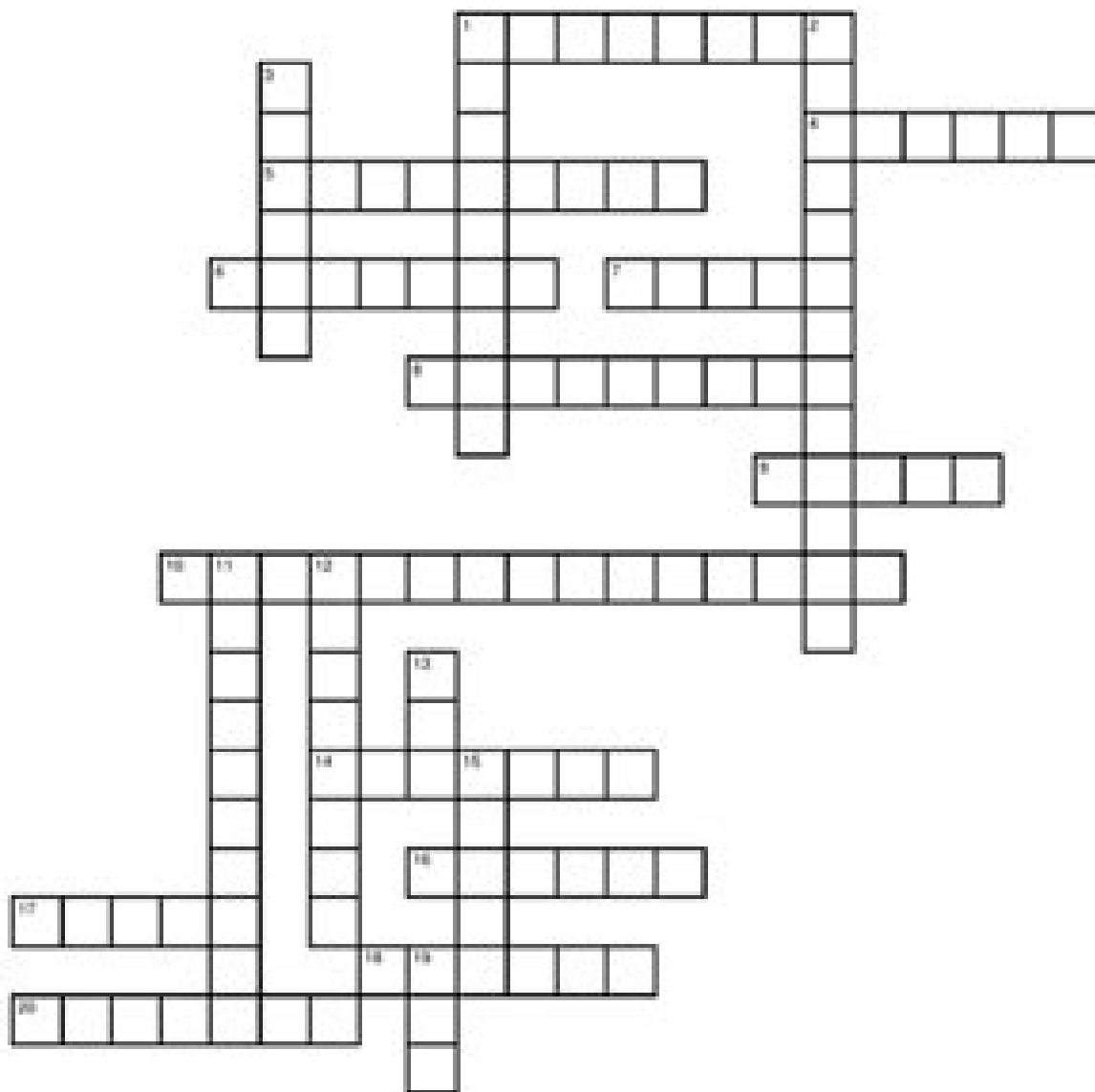


Blood Basics Review

Name _____



Across

1. Indicates the presence of a specific protein; discovered while studying Pithus monkeys
4. People with Type O blood are called universal _____ because they can give blood to anyone
5. Clotting factors in our the plasma that clot together to prevent blood loss from a wound
6. Type of blood stain created by the force of gravity an appears as drops or blood pools
7. Blood type that does not contain A or B agglutinogens
8. Type of bloodstain that is created when a force is applied to the source of the blood, such as cast-off, impact spatters, and arterial spurting
9. Our blood type is determined by these, which are inherited from our parents
10. Component of blood that is part of the our immune system and destroy pathogens in our body
14. Bloodstains created by the application of force to the area where the blood originated
15. Pointed edges of a droplet that radiate out from the spatter and can help to determine the direction of force
17. Blood type that contains B agglutinogens
18. Yellowish liquid part of the blood
20. Used by investigators to detect the presence of blood

Down

1. People with Type AB blood are called universal _____ because they can receive blood from anyone.
2. Most abundant cells in our body that are produced in the bone marrow
3. Blood type that contains both A and B agglutinogens
11. Protein found in red blood cells that carries oxygen
12. Patterns created when a bloody object comes in contact with another surface; also called contact patterns
13. Stands for bloodstain pattern analysis
15. Blood type that contains A agglutinogens
19. Test used at a crime scene to enhance a bloodstain to make it more visible in photographs

Blood Basics Review

K Payea



Blood Basics Review:

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web consider the thermochemical equation below

$$4\text{NO(g)} + 6\text{H}_2\text{O(l)} + 4\text{NH}_3\text{(g)} + 5\text{O}_2\text{(g)} \rightarrow 11\text{H}_2\text{O(l)} + 7\text{N}_2\text{(g)}$$
 a is this reaction endothermic or exothermic
 b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of H₂O

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web heats of fusion and vaporization determines what determines how much energy is required to change a substance's phase
 what is the heat of fusion formula $q = n \cdot \Delta H_{\text{fus}}$
 what is the heat of vaporization formula $q = n \cdot \Delta H_{\text{vap}}$
 what is on the x-axis of the heating curve time

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web answer a 1 explanation the entire reaction was multiplied by 2 therefore ΔH must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the ΔH per mole of O₂ and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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$$2\text{C(s)} + 3\text{O}_2\text{(g)} \rightarrow 2\text{CO}_2\text{(g)}$$
 is this endothermic or exothermic

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 $2\text{H}_2\text{(g)} + \text{O}_2\text{(g)} \rightarrow 2\text{H}_2\text{O(l)}$ energy endothermic the transfer of heat from the surroundings to the system
 $2\text{Hg(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{HgO(s)}$

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web q 140 g 4 184 J 160°C g 0°C q 9372 J 16 J 35.5 g x 1 mole KCN 0.5453 moles 65.1 g 0.5453 mole moles 5 when a 15 g sample of SrCl₂ dissolves in 80 g of water in a calorimeter the temperature drops from 42°C to 15°C calculate the ΔH for the process

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