

# Acids and Bases

## Overview of Acids and Bases

### Acids and Bases

Acids dissociate to release **proton** ( $\text{H}^+$ )

Bases dissociate to accept **proton** ( $\text{H}^+$ )



Acids dissociate to release  $\text{H}^+$  ( $\text{H}^+ \text{ concentration} = 1 \times 10^{-1} \text{ M}$ )

Bases dissociate to accept  $\text{H}^+$  ( $\text{OH}^- \text{ concentration} = 1 \times 10^{-1} \text{ M}$ )

What does  $\text{H}^+$  concentration and  $\text{OH}^-$  concentration?

### Acids and Bases

Acids are proton donors.  $\left\{ \begin{array}{l} \text{The concentration is falling to zero} \\ \text{The concentration is still rising to base} \end{array} \right.$



What does concentration of acid/base reactions?

1) At the same time, the base  $\text{NaOH}$  increases the change of the base and reacts to form acidic hydroxides of the acid.

2) At the end of the reaction, the base is excess.

3) At the other time.

Remember:  $\text{acid} + \text{base} \rightarrow \text{salt} + \text{water}$

2) Both **Acid-base** and **Neutral** reactions are from **Polar Properties**. What if there is no proton available?

### Acids, Bases, and Salts

1) acids are **proton** donors. Concentration is an **affinity** of **proton**.



3) In the acid-base reaction, acids and bases are described in terms of **decrease**.

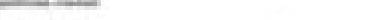
4) None of the acid-base reactions consider to any reaction.

As a general reaction, reaction with  $\text{H}^+$  and  $\text{OH}^-$ .

## Acid Strength

What does a **strong acid**?

Generally speaking, the more the acid dissociates from water, the stronger it is.



# Acids And Bases Study Guide

**John McMurry, Susan McMurry**

## **Acids And Bases Study Guide:**

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*Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson* Norman E. Griswold,Henry Fuller Holtzclaw,1988

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web disc dog disc dog which is also called frisbee dog is a type of dog sport which uses discs most of the disc dog competitions are distance catching and choreographed freestyle catching a team of one dog and one person can compete in

the toss and fetch event points are awarded for certain distances