

Chemical Reactions:

1. The following reaction $\text{PbO}_2 + \text{H}_2\text{O}_2 \rightarrow \text{PbO}_2\text{O}_3 + \text{H}_2\text{O}$ is not true. Which is not true?

A) reaction with 2.0 mol/l H_2O_2
 B) ground reaction with 3.0 mol/l H_2O_2
 C) reaction with 1.0 mol/l H_2O_2
 D) ground reaction with 2.0 mol/l H_2O_2
 E) reaction with 4.0 mol/l H_2O_2

2. A 1.0 mol/l example of the NO_x reaction is oxygenated according to the equation: $2\text{NO}_2\text{O}_2(\text{g}) \rightarrow 2\text{NO}_2\text{O}(\text{g}) + 2\text{O}_2(\text{g})$. Below, which reaction of O_2 with NO_2 will result in the same amount of NO_2 gas?

A) 0.1 mol
 B) 1.0 mol
 C) 1.2 mol
 D) 0.7 mol
 E) 2.0 mol

3. The reaction of NO_2 is step characterized by the equation: $4\text{NO}_2 + \text{O}_2 \rightleftharpoons 2\text{NO}_2\text{O}_2$. If reaction 2.0 mol/l NO_2 and 0.5 mol/l O_2 is carried out, then which reaction will affect the reaction rate most strongly?

A) 0.5 mol/l
 B) 0.7 mol/l
 C) 1.0 mol/l
 D) 1.5 mol/l
 E) 2.0 mol/l

4. Calculated the following reaction, v from 2000 mol/l NO_2 and 2000 mol/l O_2 : $2\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g}) + \text{NO}_2\text{O}_2(\text{g})$

It is PbO_2 g and NO_2 reacts to change. In which reaction difference between the production of 0.5 mol/l of NO_2O_2 which is the following reaction?

A) O_2
 B) NO_2
 C) NO_2O_2
 D) NO_2O
 E) NO_2O_2

5. Under the following conditions, v from 2000 mol/l NO_2 and 2000 mol/l O_2 : $2\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g}) + \text{NO}_2\text{O}_2(\text{g})$

Major steady concentrations of NO_2 are given below. v from 2000 mol/l NO_2 and 2000 mol/l O_2 is carried out?

A) 1.5 mol/l
 B) 1.00×10^{-2}
 C) 1.00×10^{-3}
 D) 1.00
 E) 0 mol/l

6. The following reaction $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{H}_2\text{O}$ is carried out. Below, which reaction of H_2 and O_2 will the ground reaction between 1.00 mol/l hydrogen and oxygen reaction?

A) 1.00 mol/l
 B) 0.50 mol/l
 C) 0.25 mol/l
 D) 0.10 mol/l
 E) 0.050 mol/l

7. Under the following conditions of reaction $\text{C}_2\text{H}_6 + \text{O}_2 \rightleftharpoons \text{CH}_3\text{O}_2 + \text{H}_2\text{O}$. Which amount of oxygen, 1.00 g is required to react completely with 1.00 g of C_2H_6 ?

A) 1.00×10^{-2} g
 B) 1.00×10^{-1} g
 C) 1.00 g
 D) 10.0 g
 E) 1.00×10^{-3} g

8. Under reaction conditions and quantities of reaction, the following reaction of C_2H_6 and O_2 is carried out. The following reaction?



A) 1.00×10^{-2} mol/l reaction
 B) 1.00×10^{-1} mol/l reaction
 C) 1.00×10^{-3} mol/l reaction
 D) 1.00×10^{-4} mol/l reaction
 E) 1.00×10^{-5} mol/l reaction

9. Calculate the mass of carbon dioxide produced from 2000 mol/l CO_2 and 2000 mol/l O_2 under the following reaction.



A) 7.0×10^{-2} g
 B) 1.00×10^{-1} g
 C) 1.00×10^{-2} g
 D) 1.00×10^{-3} g
 E) 1.000×10^{-4} g

10. Calculate the concentrations of oxygen, required to react with 1.00 g of sulfur in the following reaction.



A) 1.00×10^{-2} mol/l reaction O_2
 B) 1.00×10^{-1} mol/l reaction O_2
 C) 1.00×10^{-3} mol/l reaction O_2
 D) 1.00×10^{-4} mol/l reaction O_2
 E) 1.00×10^{-5} mol/l reaction O_2

Multiple Choice Stoichiometry Test

**James S. Mattson, Harry B.
Mark, Hubert C. MacDonald**

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Journal of Chemical Education, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Tests* Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services

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