

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and concentrations based on  $K_a$  and  $K_b$  and vice versa.
3. Calculate the pH of weak acids and bases.

## Vocabulary

- |                   |                             |                        |
|-------------------|-----------------------------|------------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution       |
| 4. autoionization | 5. acid rain                | 6. acid-base indicator |
|                   | 7. acid solution            | 8. pH                  |

## Key Equation

- |                                     |                                     |
|-------------------------------------|-------------------------------------|
| 1. $\text{pH} = -\log [\text{H}^+]$ | 2. $\text{pH} = -\log [\text{H}^+]$ |
| 3. $\text{pH} = -\log [\text{H}^+]$ | 4. $\text{pH} = -\log [\text{H}^+]$ |

## Part A Completion

Write the equation and/or the numerical value(s) of the unknown quantity that are indicated in the spaces provided. Be sure to include correct units when appropriate.

1. What substance is \_\_\_\_\_, a weak hydrogen ion donor?
2. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions present in a solution and equal to \_\_\_\_\_.
3. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.
4. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.
5. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.
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18. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.
19. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.
20. \_\_\_\_\_ is the concentration of  $\text{H}^+$  ions in a solution.

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# 19 2 Hydrogen Ions And Acidity Section Review Answers

**Kevin T. Patton**



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*Addison-Wesley Chemistry*, 1990 **Study Guide to Accompany Structure and Function of the Body** Linda Swisher, Gary A. Thibodeau, 2003-11 This comprehensive resource provides a variety of exercises for readers to apply and test their knowledge It contains matching fill in the blanks crossword puzzles word find unscramble the word application questions diagrams and page number references in the answer key December 2003

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