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*Schools and Academies* Louis Lafayette Williams,1890      **The Bulletin of Pharmacy** ,1913      **Journal of the New York State Teachers' Association** ,1914      **Education Outlook** ,1913      Popular Educator ,1919      **The teacher's helpful book of mental arithmetic. Standard 1(-6).** Thomas W. Piper,1881

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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$  because the mass number is the same on both sides of the equation x must have a

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web identify the missing particle in the following nuclear reaction  ${}^1_0\text{n} + {}^{235}_{92}\text{u} \rightarrow {}^{140}_{54}\text{xe} + {}^{94}_{38}\text{sr} + 2{}^1_0\text{n}$

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web answer 4 correct in the reactants and products must be the same part b what is the value of in the following nuclear reaction  ${}^{65}_{29}\text{cu} + {}^1_0\text{n} \rightarrow {}^{65}_{28}\text{ni} + {}^1_0\text{n}$  express your answer as an integer hint 1 how to approach the problem use rule 2 the total of the subscripts atomic numbers mathematically this can be written as answer

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