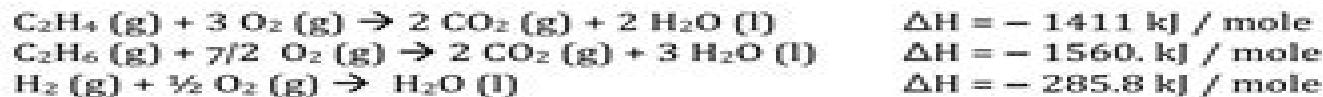


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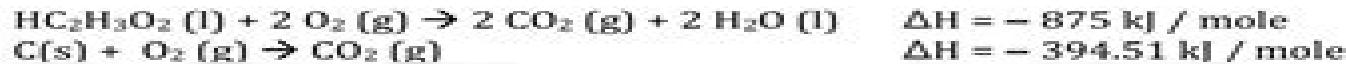
1. Calculate the  $\Delta H$  for the reaction:  $\text{C}_2\text{H}_4 \text{ (g)} + \text{H}_2 \text{ (g)} \rightarrow \text{C}_2\text{H}_6 \text{ (g)}$ .  
*from the following data:*



2. Calculate the  $\Delta H$  for the reaction:



3. Find the  $\Delta H_f^\circ$  (heat of formation) for acetic acid,  $\text{HC}_2\text{H}_3\text{O}_2$ , using the following thermochemical data:



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**Albert A Gayle**

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