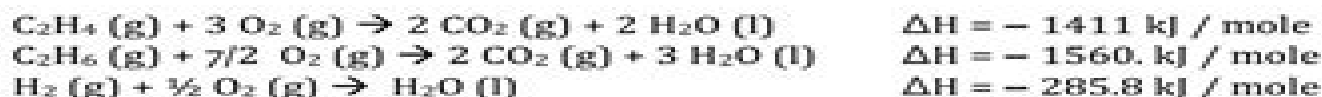


## CHEM 12 Hess's Law Worksheet

1. Calculate the  $\Delta H$  for the reaction:  $\text{C}_2\text{H}_4 (\text{g}) + \text{H}_2 (\text{g}) \rightarrow \text{C}_2\text{H}_6 (\text{g})$ ,  
from the following data:



2. Calculate the  $\Delta H$  for the reaction:



3. Find the  $\Delta H_f^\circ$  (heat of formation) for acetic acid,  $\text{HC}_2\text{H}_3\text{O}_2$ , using the following thermochemical data:



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**Albert A Gayle**



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