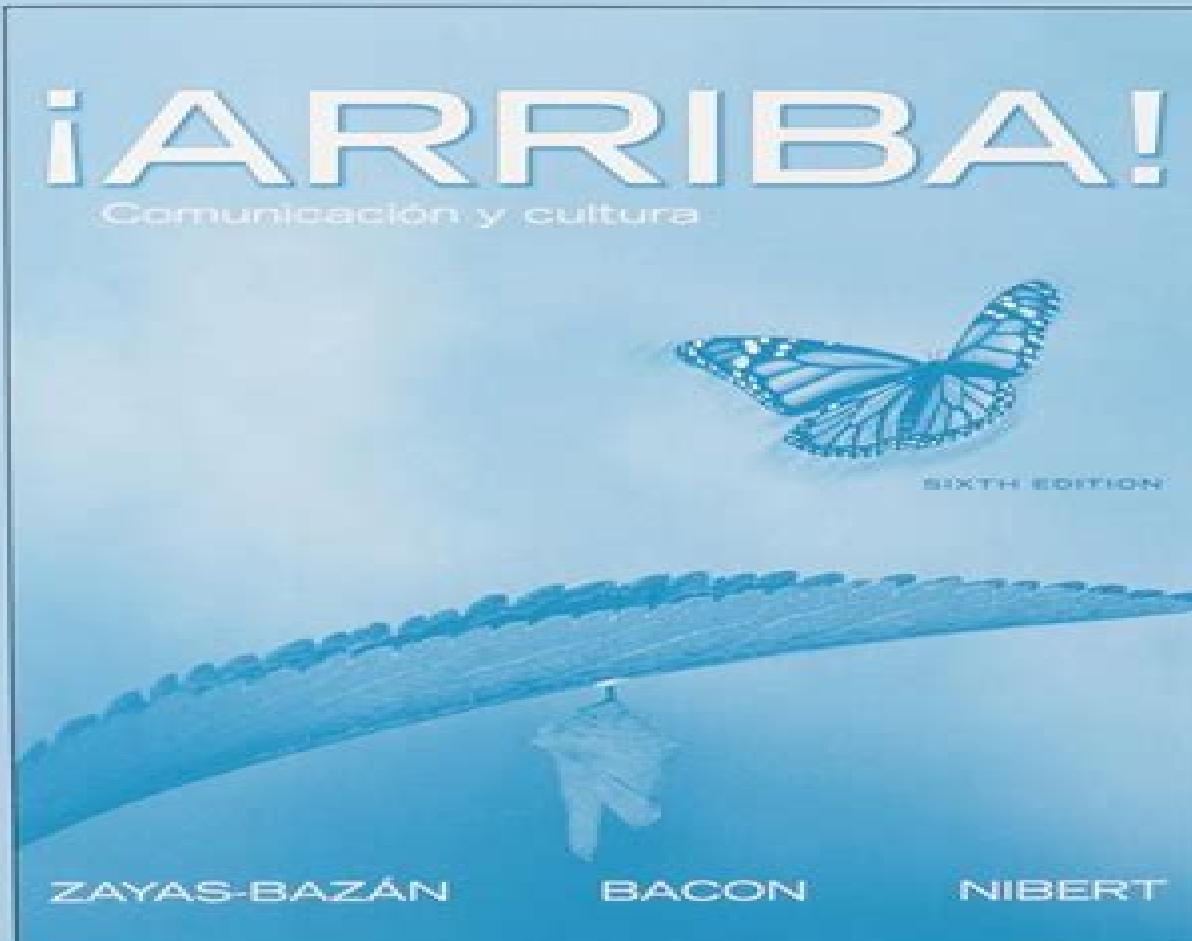


HOLLY J. NIBERT

# Student Activities Manual

*for*



PEARSON

# Arriba Student Manual

**Eduardo Zayas-bazn,Susan M.  
Bacon,Enric Figueras,Lisa Nalbone**

## **Arriba Student Manual:**

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $k_a$  or  $k_b$  from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $k_{a1}$  and  $k_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  was weighted to the nearest 0 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then ph log 0 100 1 000 ph log 0 100 1 000 b titrant volume 12 50 ml

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