

HOLLY J. NIBERT

Student Activities Manual

for

¡ARRIBA!

Comunicación y cultura



SIXTH EDITION

ZAYAS-BAZÁN

BACON

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Arriba Student Manual

**Eduardo Zayas-bazn, Susan M.
Bacon, Enric Figueras, Lisa Nalbone**

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web understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then $\text{pH} = -\log 0.100 = 1.000$ pH log 0.100 = 1.000 b titrant volume 12.50 mL

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