

1st Semester Exam
Chemistry Multiple Choice Section
(Please DO NOT write on this document)

1. What type of scientist would you consider a research project which uses previous knowledge of virus structure to find a cure for AIDS?

- a. pure scientist
- b. applied scientist
- c. biological scientist
- d. technological scientist

2. If you want to predict an experiment which will measure the effect of temperature on the properties of ice, which of the following is the **BEST** procedure for this experiment?

- a. measure 200 mL of water into 3 beakers, heat the 100 mL water to 60°C, measure the new volume after 10 minutes
- b. measure 40 mL of water into 3 beakers, heat one for 5 minutes, one for 10 minutes, and the other for 20 minutes, measure the amount of water remaining in each beaker
- c. measure 20 mL of water into 3 beakers, place the 10 beakers in 20°C, 60°C, and 100°C heating pads, heat all three beakers for 20 minutes, measure the amount of water remaining in each beaker

3. Express 540,000 in scientific notation.

- a. 544 (500)
- b. 5.44 x 10⁵
- c. 5.44 x 10⁶
- d. 5.40 x 10⁵

4. Express 0.009 x 10³ in decimal notation.

- a. 0.009
- b. 0.090
- c. 0.0009
- d. 0.00009

5. These gravity affects are objects in space.

- a. weight
- b. mass
- c. volume
- d. density
- e. energy

6. The two major characteristics of matter are that it

- a. has definite mass and shape
- b. has mass and occupies space
- c. has mass and occupies space
- d. has mass and occupies space

7. A substance which can not be broken down by chemical means is known as the _____.

- a. element
- b. compound
- c. homogeneous mixture
- d. heterogeneous mixture
- e. solution

8. The fact that solid materials are able to react relatively with acids is known as a _____.

- a. physical change
- b. physical property
- c. chemical change
- d. chemical property

9. When two metals are immersed in water, this is known as _____.

- a. physical change
- b. physical property
- c. chemical change
- d. chemical property

10. Without water the results and complications of Earth's life would be _____.

- a. another smaller planet, mass more less, therefore matter can be destroyed
- b. another smaller planet, mass more less, therefore matter can be changed into energy
- c. another smaller planet, that would weigh less than our planet, therefore matter is destroyed
- d. the planet contains matter

11. Which of the following is a good example of a heterogeneous mixture?

- a. chocolate chip cookie
- b. Bloody-Mary
- c. iron
- d. water

12. Objects with the same charge

- a. attract each other
- b. repel each other
- c. have no effect on each other
- d. not enough info

13. Which of the following is the best choice for increasing volume of a liquid?

- a. granulateductose
- b. a fruit juice
- c. beaker
- d. graduated cylinder
- e. flask

14. What should you do **BEST** if you were to spray acid all over yourself?

- a. go to the eyewash station, place your face above the sprayer, small items to use, with your eyes open
- b. stop, drop, and roll
- c. repeat the incident for your teacher
- d. not quickly under the shower head and pull the handle to turn it on

15. At the temperature of an object increases,

- a. the particles move faster
- b. the particles move more slowly
- c. it loses heat
- d. it releases cold energy

16. Metals are used to make pots and pans because they are good _____ and heat.

- a. generators
- b. conductors
- c. insulators
- d. liquids
- e. insulators

17. The best example of kinetic energy is the following list for

- a. running up stairs
- b. an electron orbiting around a nucleus
- c. an object falling on a table
- d. an atom at absolute zero

18. Which of the following is absolute zero?

- a. 0 K
- b. 0°C
- c. 0°F
- d. 0°C

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Shasha Hu

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