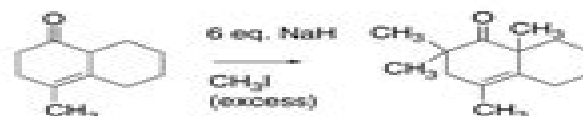


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH₂ in liquid NH₃. The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH₂ in liquid NH₃.



1. The reaction was carried out using 100% NaH and 100% NH₃ at -78°C. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was stirred for 24 hours. The reaction mixture was then poured into 100 mL of water and extracted with 100 mL of diethyl ether. The organic layer was dried over anhydrous Na₂SO₄ and concentrated under reduced pressure. The residue was purified by column chromatography on silica gel using a gradient of 0 to 100% ethyl acetate in hexanes. The pure product was obtained as a white solid.



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