

## ACIDIC AND POLYACIDIC

### Reactions with Alkalies

The term **“CARBONIC ACID**” is applied especially to acids which are weak and “polybasic” (that is, those which have more than one dissociated ionizing group, giving polybasic character). Acids with two such groups are said to have **“DIBASIC** character. Those acids which have only one ionizing group are said to have a **“MONOBASIC** character. The term **“POLYACIDIC**” is applied to acids which have more than one ionizing group, giving polyacid character.

### Reactions with Acids

- 1. **Acidic and monobasic acids** – (Acidic acids are those that have only one ionizable group, giving monobasic character. These acids have one ionizable group and the acid will not have sufficient protons to form polybasic character. They are polyacidic acids and are strong acids.)
  - a. **“Strong” acids** – These acids are the strongest and easiest to dissociate. These acids are polybasic acids and dissociate into polybasic polyacidic acids. Strong acids dissociate into monobasic acids and monobasic acids are strong acids.
  - b. **“Weak” acids** – These acids are dissociated without giving monobasic acids. These acids are polyacidic acids and are strong acids.
- 2. **Monobasic acids – monobasic acids** – These acids are dissociated without giving monobasic acids. These acids are dissociated to give monobasic acids and the acid will not have sufficient protons to form polybasic acids. These acids are polyacidic acids and are strong acids.
- 3. **Monobasic acids – polyacidic acids** – These acids are dissociated without giving monobasic acids. These acids are dissociated to give monobasic acids and the acid will not have sufficient protons to form polybasic acids. These acids are polyacidic acids and are strong acids.
- 4. **Polyacidic acids – polyacidic acids** – These acids are dissociated without giving monobasic acids. These acids are dissociated to give monobasic acids and the acid will not have sufficient protons to form polybasic acids. These acids are polyacidic acids and are strong acids.
- 5. **Monobasic acids – polyacidic acids** – These acids are dissociated without giving monobasic acids. These acids are dissociated to give monobasic acids and the acid will not have sufficient protons to form polybasic acids. These acids are polyacidic acids and are strong acids.

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**DP Hallahan**

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