



# EARTH SCIENCE

## INTRODUCTION

### COMPONENTS OF STUDY

- Geology: Earth physical & historical
- Metgeology: Weather & Climate
- Oceanography: The oceans
- Astronomy: What is external to the atmosphere

### THE FOUR SPHERES (See The Water Cycle)

- Lithosphere: The solid earth
- Hydrosphere: The liquid earth
- Atmosphere: The gaseous earth
- Biosphere: Where life exists

### CONCERN IN OUR PHYSICAL ENVIRONMENT

- Resources: Water, soil, metallic & non-metallic elements, sunlight, sunlight, wind, etc.
- The capita use of metallic and non-metallic elements is 11 tons per year
- Renewable vs. non-renewable resources
- Environmental and population problems

## MINERALS

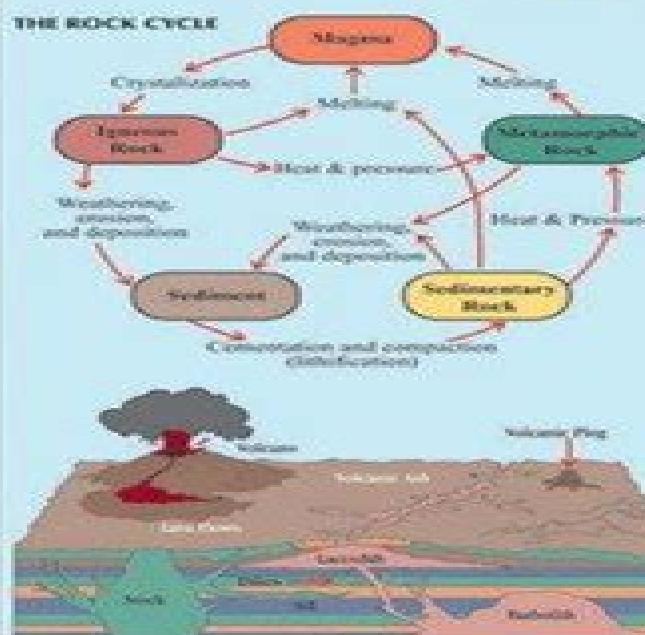
- A mineral is a naturally occurring, inorganic, solid material with a definite atomic structure.
- An ore is a useful mineral resource.
- A. Atom and Minerals**
  - Atoms in ancient Greek means "un-cut" i.e., the Greeks believed the atom to be the smallest particle of matter.
  - Atoms are composed of neutrons, protons and electrons.
  - Atomic number: Number of protons in nucleus.
  - Atomic weight: Average weight of an atom.
  - Isotopes: Atoms with the same atomic number but different numbers of neutrons in the nucleus.
  - Forces: Strong, weak, electromagnetic and gravity.
- Properties of Minerals**
  - Color: Appearance or quality of light from surface.
  - Texture: Nature of light and cause of color.
  - Streak: Color of material in powdered form.
  - HARDNESS: Mohs' Scale (See Figure Below).
  - Scratch Test: Higher numbered materials can scratch lower numbered materials.
  - Shape is determined by cleavage, crystal form and structure.
  - Specific Gravity:
    - The ratio of the weight of a mineral to the weight of an equal volume of water.
    - Density of water = 1 gram/cubic centimeter.
    - Iron, iron = 7.7, aluminum = 2.7, aluminum = 2.2.

MOM'S SCALE		MINERALS
10	Diamond	
9	Corundum	
8	Topaz	
7	Quartz	
6	feldspar	
5	Spatite	
4	Fluorite	
3	Gypsum	
2	Orthoclase	
1	Halite	
0	Table	

### C. Mineral Concepts

- Silicates: Minerals with silicon and oxygen.
- Silicates are a large class which forms a pyramidal-shaped structure with oxygen.
- Other forms include simple non-silicate minerals, oxides, sulfides, noble elements, 2-3 element hydrocarbons and quartz.
- Non-Silicates
- Make up one-fourth of continental crust.
- Carbides: Minerals with carbon and oxygen including calcium, those which are present limestone (calcium) and marble (calcium carbonate).
- Oxides: Oxygen-banded solids, i.e., iron sulfides (FeS), sulfates (MgSO<sub>4</sub>), halides (CaCl<sub>2</sub>, F).
- Halite: Mineral form of salt.
- Corundum: Plasma, calcium.
- Noble metals: Iron, zinc, gold, silver, nickel.

## THE ROCK CYCLE



## ROCK TYPES

### METAMORPHIC ROCKS

#### A. Formation

- Metamorphism**: To change form.
- Metamorphic rock**: Material under intense pressure, high temperatures.
- Contact metamorphism**: Metamorphism caused by proximity to magma.
- Low-grade metamorphism**: Shallow metamorphism.
- High-grade metamorphism**: Deeper metamorphism.
- foliation**: Layers during metamorphism.
- Metamorphic rocks are harder than sedimentary rocks.**
- Metamorphism changes texture.**
- Low-grade metamorphism makes materials more compact, dense, altered.**
- High-grade metamorphism makes materials more compact, dense, altered.**
- Metamorphic rocks form with one another.**

#### B. Agents of Metamorphism

- Heat**
  - most important agent
  - possible energy for chemical reactions
  - For example, clay converts into a mineral at great temperature
- Pressure**
- Chemical Activity**: Most common chemical agent in nature.

#### C. Types of Metamorphic Rocks

- Metamorphic**: Fine-grained, dark, shiny rock with randomly arranged minerals.
- Schist**: Fine-grained, often grey, foliated rock with many shiny cleavage planes of mica minerals aligned by pressure.
- Migmatite**: Granular or mica-schist rock.
- Phyllite**: Shiny, foliated rock more compactly foliated than schist.
- Slate**: Foliated rock, more compactly foliated and w/ higher metamorphic grade than phyllite.
- Gneiss**: Foliated, banded rock w/ the highest metamorphic grade.
- Quartzite**: Very hard, granular quartz rock, formed from sandstone.

### ROCKS' REACTION SERIES

It is a mineral, which has already formed, reacts w/ other minerals, it will react w/ the remaining magma to produce the next mineral in the sequence. For example, olivine reacts w/ silica, which reacts w/ remaining magma to form pyroxene.

#### D. Magma and other Igneous Rocks

- Calcsilicate rocks: w/ other minerals, form small crystals w/ foliation.
- Tholeiitic rocks: w/ sodium in the magma to form olivine and more sodium-rich.

#### E. End of Cooling

- Without anything to allow cool, remaining magma w/ more high-melting content, and quartz w/ it.
- Without cooling to complete, anything that cooled at the same time will usually be the same for one another (quartz, feldspar, and quartz) cool down over time for major growth.



# Earth Science High School Essential Question

**Ohio. Department of Education**

## **Earth Science High School Essential Question:**

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