

Chemistry Final Exam Review

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- ____ 1. The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry?
a. bio c. physical
b. inorganic d. analytical
- ____ 2. Which of the following is NOT an example of matter?
a. air c. smoke
b. heat d. water vapor
- ____ 3. A golf ball has more mass than a tennis ball because it _____.
a. takes up more space c. contains different kinds of matter
b. contains more matter d. has a definite composition
- ____ 4. An example of an extensive property of matter is _____.
a. temperature c. mass
b. pressure d. hardness
- ____ 5. All of the following are physical properties of matter EXCEPT _____.
a. mass c. melting point
b. color d. ability to rust
- ____ 6. Which state of matter has a definite volume and takes the shape of its container?
a. solid c. gas
b. liquid d. both b and c
- ____ 7. Which of the following CANNOT be classified as a substance?
a. table salt c. nitrogen
b. air d. gold
- ____ 8. Which of the following is a heterogeneous mixture?
a. air c. steel
b. salt water d. soil
- ____ 9. Which of the following is a heterogeneous mixture?
a. vinegar in water c. oil and vinegar
b. milk d. air
- ____ 10. Separating a solid from a liquid by evaporating the liquid is called _____.
a. filtration c. solution
b. condensation d. distillation
- ____ 11. A substance that can be separated into two or more substances only by a chemical change is a(n) _____.
a. solution c. mixture
b. element d. compound
- ____ 12. The chemical symbol for iron is _____.
a. fe c. Fe
b. FE d. Ir

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research in the first two years How to design appropriate course based research for first and second year students How to mentor a class rather than individual students How students can disseminate the results of their research Possible citizen science projects appropriate for the first and second years Providing additional resources available to support course based research in the first two years Designed for faculty at four year and two year colleges and including examples from the sciences the social sciences and the humanities the strategies and methods described can be adapted to disciplines not specifically mentioned in the book Many faculty are hesitant to engage first and second year students in undergraduate research because they worry students don't know enough to conduct authentic research in their discipline because they worry about the time it will take to develop activities for these students and because they wonder how they can mentor a whole class of students doing research The authors have successfully dealt with these issues and provide examples of how it is done

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