

# Unit Test with answer **EDITABLE**

**Chapter 1: Review Questions**

1. Indicate which of the following numbers is accurate, precise, and both accurate and precise.

- 1.0000000000000001
- 1.0000000000000001
- 1.0000000000000001
- 1.0000000000000001

2. Estimate the following quantities with the correct number of significant digits.

- $1.000 \times 10^3 \text{ g} = 1.000 \times 10^3 \text{ g}$
- $10.000 \times 10^3 \text{ g} = 10.000 \times 10^3 \text{ g}$
- $1.00 \times 10^3 \text{ g} = 1.00 \times 10^3 \text{ g}$
- $1.000 \times 10^3 \text{ g} = 1.000 \times 10^3 \text{ g}$

3. The following data are collected as four different students work independently to measure the mass of a pencil eraser. They then add the data.

Student	Mass of eraser (g)
1	0.0000000000000001
2	0.0000000000000001
3	0.0000000000000001
4	0.0000000000000001

Which one of these has high accuracy but low precision?

- None of the data has both accuracy and precision.
- None of the data has high accuracy but low precision.
- None of the data has good precision but no accuracy.



4. A graduated cylinder is containing 10.0 mL of water. When 10.0 mL of 10% (w/v) copper sulphate solution is added to the cylinder, the total volume of the solution is 20.0 mL. The density of copper sulphate is 1.0 g/mL.

5. A graduated glass jar holds 10.0 mL of water. When 10.0 mL of 10% (w/v) copper sulphate solution is added to the jar, the total volume of the solution is 20.0 mL. The density of copper sulphate is 1.0 g/mL.

6. A graduated cylinder contains 10.0 mL of water. A 10.0 mL graduated cylinder containing 10.0 mL of copper sulphate solution is then added to the graduated cylinder. The density of copper sulphate is 1.0 g/mL. The total volume of the solution in the graduated cylinder is 20.0 mL.

7. A graduated cylinder contains 10.0 mL of water. A 10.0 mL graduated cylinder containing 10.0 mL of copper sulphate solution is then added to the graduated cylinder. The density of copper sulphate is 1.0 g/mL. The total volume of the solution in the graduated cylinder is 20.0 mL.

# AP Chemistry

# Apex Chemistry Unit 3 Test

**Siddappa N.Byrareddy**

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**Resources in Education** ,1972-10 **The New American Encyclopedic Dictionary** Robert Hunter,Edward Thomas Roe,Le Roy Hooker,Thomas W. Handford,1906 **Engineering Journal** ,1924 **The American Dictionary and Cyclopedias** Robert Hunter,1900 The Wireless World and Radio Review ,1925

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