

Chemical Reactions:

1. The following reaction $\text{PbO}_2 + \text{H}_2\text{O}_2 \rightarrow \text{PbO}_2\text{O}_3 + \text{H}_2\text{O}$ is not true. Which is not true?

A) reaction with 2.0 mol/l H_2O_2
 B) ground reaction with 3.0 mol/l H_2O_2
 C) reaction with 1.0 mol/l H_2O_2
 D) ground reaction with 2.0 mol/l H_2O_2
 E) reaction with 4.0 mol/l H_2O_2

2. A 1.0 mol/l example of the NO_x reaction is oxygenated according to the equation: $2\text{NO}_2\text{O}_2(\text{g}) \rightarrow 2\text{NO}_2\text{O}(\text{g}) + 2\text{O}_2(\text{g})$. Below, which reaction of O_2 with NO_2 will result in the same amount of NO_2 gas?

A) 0.1 mol
 B) 1.0 mol
 C) 1.2 mol
 D) 0.7 mol
 E) 2.0 mol

3. The reaction of NO_2 is step characterized by the equation: $4\text{NO}_2 + \text{O}_2 \rightleftharpoons 2\text{NO}_2\text{O}_2$. If reaction 1.0 mol/l NO_2 and 0.2 mol/l O_2 is carried out, then which reaction will affect the reaction rate most?

A) 0.5 mol/l
 B) 0.7 mol/l
 C) 1.0 mol/l
 D) 1.2 mol/l
 E) 2.0 mol/l

4. Calculated on the following reaction, v from 2000 mol/l NO_2 and 1000 mol/l O_2 is?

$\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g})$

$\text{v} = \text{P}_2 \cdot \text{g} \cdot \text{mol}^{-1} \cdot \text{O}_2$ is the reaction rate of O_2 , v is the reaction rate of NO_2 , P_2 is the pressure in 1.0 mol/l of NO_2 , g is the reaction rate of NO_2O , $\text{mol}^{-1} \cdot \text{O}_2$ is the reaction rate of O_2 .

A) 0.0001
 B) 0.0002
 C) 0.0003
 D) 0.0004
 E) 0.0005

5. Calculated on the following reaction, v from 2000 mol/l NO_2 and 1000 mol/l O_2 is?

$\text{NO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{NO}_2\text{O}(\text{g})$

$\text{v} = \text{P}_2 \cdot \text{g} \cdot \text{mol}^{-1} \cdot \text{O}_2$ is the reaction rate of O_2 , v is the reaction rate of NO_2 , P_2 is the pressure in 1.0 mol/l of NO_2 , g is the reaction rate of NO_2O , $\text{mol}^{-1} \cdot \text{O}_2$ is the reaction rate of O_2 .

A) 0.0001
 B) 0.0002
 C) 0.0003
 D) 0.0004
 E) 0.0005

6. The following reaction $2\text{H}_2\text{O}_2 \rightarrow 2\text{H}_2\text{O} + \text{O}_2$ is not true. Which reaction of H_2O_2 will result in the same amount of hydrogen and oxygen?

A) 1.0 mol/l
 B) 2.0 mol/l
 C) 0.5 mol/l
 D) 0.1 mol/l
 E) 0.2 mol/l

7. Under the following conditions of reaction, which amount of oxygen O_2 is required to react completely with 1.0 g of CH_3O_2 ?

A) 0.0001×10^{22} g
 B) 0.001×10^{22} g
 C) 0.005×10^{22} g
 D) 0.01×10^{22} g
 E) 0.1×10^{22} g

8. Under reaction conditions and quantities of reaction, which amount of O_2 is required to react completely with 1.0 g of CH_3O_2 ?



A) 0.0001×10^{22} molecules
 B) 0.001×10^{22} molecules
 C) 0.005×10^{22} molecules
 D) 0.01×10^{22} molecules
 E) 0.1×10^{22} molecules

9. Calculate the mass of carbon dioxide produced from 10.0 g of octane, C_8H_{18} , in the following reaction.



A) 7.0×10^{-3} g
 B) 1.7×10^{-3} g
 C) 0.01×10^{-3} g
 D) 1.2×10^{-3} g
 E) 1.8×10^{-3} g

10. Calculate the mass and percentage of oxygen required to react with 1.0 g of sulfur in the following reaction.



A) 0.0001×10^{22} molecules O_2
 B) 0.001×10^{22} molecules O_2
 C) 0.005×10^{22} molecules O_2
 D) 0.01×10^{22} molecules O_2
 E) 0.1×10^{22} molecules O_2

Multiple Choice Stoichiometry Test

Paul Foglino, Princeton Review (Firm)

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Cambridge Checkpoints VCE Chemistry Units 3 and 4 2013 Roger Slade, Maureen Slade, 2012-12-14 Cambridge Checkpoints VCE are updated regularly to provide you with the most up to date exam preparation available **Cracking the AP Chemistry Exam, 2012 Edition** Paul Foglino, 2011-08-02 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests *Roadmap to the Regents* Sasha Alcott, 2003 If Students Need to Know It It's in This Book This book develops the chemistry skills of high school students It builds skills that will help them succeed in school and on the New York Regents Exams Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests Each year we help more than 2 million students score higher and earn better grades We Know the New York Regents Exams Our experts at The Princeton Review have analyzed the New York Regents Exams and this book provides the most up to date thoroughly researched practice possible We break down the test into individual skills to familiarize students with the test's structure while increasing their overall skill level We Get Results We know what it takes to succeed in the classroom and on tests This book includes strategies that are proven to improve student performance We provide a breakdown of the skills based on New York standards and objectives hundreds of practice questions organized by skill two complete practice New York Regents Exams in Physical Setting Chemistry *Master the CLEP* Peterson's, 2012-04-30 Peterson's Master the CLEP is the essential prep guide for non traditional students who are looking to begin a college career and for current students who want to save on tuition costs by testing out of certain courses When you use this guide you will get a basic overview of the CLEP a look at the general examinations and the vast number of subject specific exams tips for registering and preparing for the CLEP and a guide to interpreting your test scores This is followed by extensive reviews of the five CLEP general examinations with pre and post tests and subject matter overviews that will help you pinpoint your strengths and weaknesses The Appendix will go into detail about a number of CLEP subject exams *CLEP Success* Peterson's, 2007-05-04 Offers advice on improving scores on the CLEP and includes reviews for five CLEP subjects and five full length practice tests with explanatory answers

Journal of Chemical Education, 1926 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers *Tests* Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services

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Computer-assisted Instruction in Chemistry: General approach James S. Mattson,Harry B. Mark,Hubert C. MacDonald,1974 **Proceedings of Conference on Computers in the Undergraduate Curricula ,1973**

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