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# Applied Petroleum Reservoir Engineering

## 3rd Edition Solution Manual

### CHAPTER 2

2.1 using Equation 2.4:  $pV = nRT$

$$(a) \quad V = \frac{1(10.73)(60 + 460)}{14.7} = 379.6 \text{ SCF}$$

$$(b) \quad V = \frac{1(10.73)(32 + 460)}{14.7} = 359.1 \text{ SCF}$$

$$(c) \quad V = \frac{1(10.73)(80 + 460)}{14.7 + 10/16} = 378.1 \text{ SCF}$$

$$(d) \quad V = \frac{1(10.73)(60 + 460)}{15.025} = 371.4 \text{ SCF}$$

2.2 (a) number of moles of methane =  $10/16 = .625$  moles

number of moles of ethane =  $20/30 = .667$  moles

total number of moles =  $.625 + .667 = 1.292$  moles

(b) from Equation 2.4

$$p = \frac{1.292(10.73)(550)}{500} = 15.25 \text{ psia}$$

# Applied Reservoir Engineering Solution Manual

**Ensheng Dong**



## **Applied Reservoir Engineering Solution Manual:**

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**Reviews**, 1948      *The Publishers' Trade List Annual*, 1886      *Scientific and Technical Books and Serials in Print*, 1984

*Scientific and Technical Books in Print*, 1972      **Water Supply Engineering** Harold Eaton Babbitt, James Joseph Doland, 1929      The Engineering Record, Building Record and the Sanitary Engineer, 1901      **Engineering Record, Building Record and Sanitary Engineer**, 1886      *Fire and Water Engineering*, 1920      **Subject Guide to Books in Print**, 1991      *The Journal of Canadian Petroleum Technology*, 1997      **The Log Analysis Handbook: Quantitative log analysis methods** E. R. Crain, 1986      **Journal of Industrial and Engineering Chemistry**, 1909      **The Journal of Industrial and Engineering Chemistry**, 1910      **Applied Hydraulics**, 1957

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## **Table of Contents Applied Reservoir Engineering Solution Manual**

1. Understanding the eBook Applied Reservoir Engineering Solution Manual
  - The Rise of Digital Reading Applied Reservoir Engineering Solution Manual
  - Advantages of eBooks Over Traditional Books
2. Identifying Applied Reservoir Engineering Solution Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Applied Reservoir Engineering Solution Manual
  - User-Friendly Interface

4. Exploring eBook Recommendations from Applied Reservoir Engineering Solution Manual
  - Personalized Recommendations
  - Applied Reservoir Engineering Solution Manual User Reviews and Ratings
  - Applied Reservoir Engineering Solution Manual and Bestseller Lists
5. Accessing Applied Reservoir Engineering Solution Manual Free and Paid eBooks
  - Applied Reservoir Engineering Solution Manual Public Domain eBooks
  - Applied Reservoir Engineering Solution Manual eBook Subscription Services
  - Applied Reservoir Engineering Solution Manual Budget-Friendly Options
6. Navigating Applied Reservoir Engineering Solution Manual eBook Formats
  - ePub, PDF, MOBI, and More
  - Applied Reservoir Engineering Solution Manual Compatibility with Devices
  - Applied Reservoir Engineering Solution Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Applied Reservoir Engineering Solution Manual
  - Highlighting and Note-Taking Applied Reservoir Engineering Solution Manual
  - Interactive Elements Applied Reservoir Engineering Solution Manual
8. Staying Engaged with Applied Reservoir Engineering Solution Manual
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Applied Reservoir Engineering Solution Manual
9. Balancing eBooks and Physical Books Applied Reservoir Engineering Solution Manual
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Applied Reservoir Engineering Solution Manual
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Applied Reservoir Engineering Solution Manual
  - Setting Reading Goals Applied Reservoir Engineering Solution Manual
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Applied Reservoir Engineering Solution Manual
  - Fact-Checking eBook Content of Applied Reservoir Engineering Solution Manual
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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