

1st Semester Exam
Chemistry Multiple Choice Section
(Please DO NOT write on this document)

1. What type of scientist would you consider a research project which uses previous knowledge of virus structure to find a cure for AIDS?
- a. pure scientist
 - b. applied scientist
 - c. biological scientist
 - d. technological scientist
2. If you want to predict an experiment which will measure the effect of temperature on the properties of ice, which of the following is the **BEST** procedure for this experiment?
- a. measure 200 mL and 500 mL of water into 3 beakers, heat the 100 mL water to 60°C, measure the new volume after 10 minutes
 - b. measure 400 mL of water into 3 beakers, heat one for 5 minutes, one for 10 minutes, and the other for 20 minutes, measure the amount of water remaining in each beaker
 - c. measure 200 mL of water into 3 beakers, place the 100 mL ice at 20°C, 60°C, and 110°C heating pads, heat all three beakers for 20 minutes, measure the amount of water remaining in each beaker
3. Express $541,000$ in scientific notation.
- a. 541×10^3
 - b. 5.41×10^5
 - c. 5.41×10^6
 - d. 541×10^7
4. Express 0.009×10^3 in decimal notation.
- a. 0.009
 - b. 0.09
 - c. 0.0009
 - d. 0.00009
5. These gravity affects are objects in space.
- a. weight
 - b. mass
 - c. volume
 - d. density
 - e. energy
6. The two major characteristics of matter are that it
- a. has definite mass and shape
 - b. has mass and occupies space
 - c. has mass and occupies space
 - d. has mass and occupies space
 - e. has mass and occupies space
7. A substance which can not be broken down by chemical means is known as the _____.
- a. element
 - b. compound
 - c. homogeneous mixture
 - d. heterogeneous mixture
 - e. solution
8. The fact that solid materials are able to react relatively with acids is known as a
- a. physical change
 - b. physical property
 - c. chemical change
 - d. chemical property
9. When two metals are immersed in water, this is known as
- a. physical change
 - b. physical property
 - c. chemical change
 - d. chemical property
10. Without water the results and complications of the following is/are **Corrosion**.
- a. another metal is lost, mass stays the same, therefore matter can be destroyed
 - b. another metal is lost, mass stays the same, therefore matter can be changed into energy
 - c. another metal is lost, mass stays the same at the original weight, therefore matter is destroyed
 - d. the other contains matter
11. Which of the following is a good example of a homogeneous mixture?
- a. chocolate chip cookie
 - b. Bloody-Mary
 - c. trees
 - d. water
12. Objects with the same charge
- a. attract each other
 - b. repel each other
 - c. have no effect on each other
 - d. not enough info
13. Which of the following is the best choice for increasing volume of a liquid?
- a. granulateductose
 - b. a fruit juice
 - c. beaker
 - d. graduated cylinder
 - e. flask
14. What should you do **FIRST** if you were to spray acid off your pants?
- a. go to the eyewash station, place your face above the sprayer, pull pants down, with your eyes open
 - b. stop, sit-up, and roll
 - c. repeat the incident for your instructor
 - d. not quickly under the shower head and pull the handle to turn it on
15. At the temperature of an object increases,
- a. the particles move slower
 - b. the particles move more slowly
 - c. the particles move faster
 - d. the particles move faster
 - e. the particles move more slowly
16. Metals are used to make pots and pans because they are good _____ and heat.
- a. generators
 - b. conductors
 - c. insulators
 - d. liquids
 - e. insulators
17. The best example of kinetic energy is the following list for
- a. running up stairs
 - b. an electron orbiting around a nucleus
 - c. an object falling on a table
 - d. an object at absolute zero
18. Which of the following is absolute zero?
- a. 0 K
 - b. 0°C
 - c. 0°F
 - d. 0°C

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