

Name: \_\_\_\_\_

Period: \_\_\_\_\_

### Atoms and bonding study guide

- 1) What is the force that holds atoms together?
- 2) Why do atoms form chemical bonds?
- 3) Name the following compounds:  $K_2O$ ,  $NaF$ ,  $PI_3$
- 4) What are the subatomic particles found in the outermost energy level of an atom?
- 5) How many valence electrons do most atoms need to be stable? \_\_\_\_\_
- 6) Which elements only require 2 valence electrons to be stable?
- 7) Draw an electron dot diagram for: He, Li, C, Ca, Al, Si, P, O, Cl
- 8) Which of the above elements are likely to bond with one another (put them in pairs, one will be left out)?
- 9) Write the charges for the following ions: H, Li, K, Cl, Si
- 10) Write the correct chemical formula for the following covalent compounds: Dihydrogen disulfide, Carbon disulfide, Carbon tetraiodide
- 11) Write the correct chemical formula for the following ionic compounds: Magnesium nitride, Sodium chloride, Silicon oxide

# Atoms And Bonding Assessment Study Guide

**Arshad Iqbal**



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