

Atomic Structure - Study Guide

1. Identify the location of the protons, neutrons, and electrons.

Protons & neutrons are located in the nucleus (center) and have positive charges.

Electrons are located throughout the atom, and have negative charges.

Protons and neutrons have mass, while electrons have little mass.



2. Define the three subatomic particles and their charges.

Subatomic Particle	Symbol	Charge
Proton	p^+	positive
Neutron	n^0	neutral
Electron	e^-	negative

Mass number = the sum of the number of protons and neutrons in the nucleus.

Atomic number = the number of protons in the nucleus.

3. Define the three subatomic particles and their charges.

Protons and neutrons are located in the nucleus, and have positive charges. Electrons are located throughout the atom, and have negative charges.

4. Define the three subatomic particles and their charges.

Protons and neutrons are located in the nucleus, and have positive charges. Electrons are located throughout the atom, and have negative charges.

Any two protons have repulsion.

5. Define the three subatomic particles and their charges.

Protons and neutrons are located in the nucleus, and have positive charges. Electrons are located throughout the atom, and have negative charges.

Protons and neutrons have mass, while electrons have little mass.

Subatomic Particle	Symbol	Charge	Mass
Proton	p^+	positive	large
Neutron	n^0	neutral	large

Electron
Symbol
 e^-

Li

Mass = 7

Atomic Structure Study Guide And Nuclear Processes

SA Dillow

Atomic Structure Study Guide And Nuclear Processes:

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