

## 14.3

## HYDROGEN IONS AND ACIDITY

## Section Review

## Objectives

1. Calculate concentrations based on the pH of a given acid or hydrogen ion concentration.
2. Calculate hydrogen ion concentrations from  $\text{pH}$  and  $\text{pOH}$ , and vice versa from  $\text{pH}$  and  $\text{pOH}$ .
3. Calculate the  $\text{pH}$  of a weak acid.

## Vocabulary

- |                   |                               |                  |
|-------------------|-------------------------------|------------------|
| 1. autoionization | 2. hydrogen ion concentration | 3. acid solution |
| 4. autoionization | 5. $\text{pH}$                | 6. $\text{pOH}$  |

## Key Equation

- |                                    |                                  |
|------------------------------------|----------------------------------|
| 1. $\text{pH} = -\log[\text{H}^+]$ | 2. $\text{pH} = 14 - \text{pOH}$ |
| 3. $\text{pH} = -\log[\text{H}^+]$ | 4. $\text{pH} = 14 - \text{pOH}$ |

## Part A Completion

Read the statement and fill in the blank with the word or phrase that is indicated in parentheses. Write the word or phrase in the space provided.

1. When a substance ionizes, it releases hydrogen ions. (acidic)
2. The concentration of  $\text{H}^+$  in the autoionization of water is  $1.0 \times 10^{-7}$  moles per liter at 25°C. (neutral)
3. The  $\text{pH}$  of a solution is defined as the negative logarithm of the concentration of  $\text{H}^+$ . (acidic)
4. The  $\text{pH}$  of a solution is defined as the negative logarithm of the concentration of  $\text{H}^+$ . (acidic)
5. The  $\text{pH}$  of a solution is defined as the negative logarithm of the concentration of  $\text{H}^+$ . (acidic)
6. The  $\text{pH}$  of a solution is defined as the negative logarithm of the concentration of  $\text{H}^+$ . (acidic)
7. The  $\text{pH}$  of a solution is defined as the negative logarithm of the concentration of  $\text{H}^+$ . (acidic)

1. **acidic**

2.  **$1.0 \times 10^{-7}$**

3. **acidic**

4. **acidic**

5. **acidic**

6. **acidic**

7. **acidic**

8. **acidic**

9. **acidic**

10. **acidic**

11. **acidic**

# 19 2 Hydrogen Ions And Acidity Answer Key

**Eileen Ramsden**



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### Table of Contents 19 2 Hydrogen Ions And Acidity Answer Key

1. Understanding the eBook 19 2 Hydrogen Ions And Acidity Answer Key
  - The Rise of Digital Reading 19 2 Hydrogen Ions And Acidity Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying 19 2 Hydrogen Ions And Acidity Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 19 2 Hydrogen Ions And Acidity Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from 19 2 Hydrogen Ions And Acidity Answer Key
  - Personalized Recommendations
  - 19 2 Hydrogen Ions And Acidity Answer Key User Reviews and Ratings
  - 19 2 Hydrogen Ions And Acidity Answer Key and Bestseller Lists

5. Accessing 19 2 Hydrogen Ions And Acidity Answer Key Free and Paid eBooks
  - 19 2 Hydrogen Ions And Acidity Answer Key Public Domain eBooks
  - 19 2 Hydrogen Ions And Acidity Answer Key eBook Subscription Services
  - 19 2 Hydrogen Ions And Acidity Answer Key Budget-Friendly Options
6. Navigating 19 2 Hydrogen Ions And Acidity Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - 19 2 Hydrogen Ions And Acidity Answer Key Compatibility with Devices
  - 19 2 Hydrogen Ions And Acidity Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 19 2 Hydrogen Ions And Acidity Answer Key
  - Highlighting and Note-Taking 19 2 Hydrogen Ions And Acidity Answer Key
  - Interactive Elements 19 2 Hydrogen Ions And Acidity Answer Key
8. Staying Engaged with 19 2 Hydrogen Ions And Acidity Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 19 2 Hydrogen Ions And Acidity Answer Key
9. Balancing eBooks and Physical Books 19 2 Hydrogen Ions And Acidity Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 19 2 Hydrogen Ions And Acidity Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 19 2 Hydrogen Ions And Acidity Answer Key
  - Setting Reading Goals 19 2 Hydrogen Ions And Acidity Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 19 2 Hydrogen Ions And Acidity Answer Key
  - Fact-Checking eBook Content of 19 2 Hydrogen Ions And Acidity Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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