

The University of
Akron/Summit County
Sheriff's Office



POLICE ACADEMY

ENTRY REQUIREMENTS

- Must be 20-1/2 at time of application
- Must be in good physical condition
- Must have a valid driver's license
- Must not have any Felony or arrests for Domestic Violence



Academy Dates:

Full-time Day Academy

July 31, 2023 to Nov. 27, 2023
Mon-Sat, 8:00-4:30 pm

Part-time Evening Academy

July 31, 2023 to April 12, 2024
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USAF Formal Schools United States. Department of the Air Force,1987 USAF Formal Schools United States. Dept. of the Air Force,1987 *Regulations Governing the Admission of Candidates Into the Naval Academy* United States Naval Academy,1918 *The Academy* ,1887 **Bulletin** Institute of International Education (New York, N.Y.),1925 **School and Society** ,1922 **School & Society** ,1922 **School & Society** James McKeen Cattell,Will Carson Ryan,Raymond Walters,1922 **Annual Bulletin of Colorado College and Cutler Academy** Colorado College,1896 **Fellowships and Scholarships Open to American Students for Study in Foreign Countries** Institute of International Education (New York, N.Y.),1929 **Regulations of the United States Naval Academy** United States Naval Academy,1929
Brugger V. Joseph Academy, Inc ,2001 The Annals of the American Academy of Political and Social Science ,1917
Bulletin [of] the Institute of International Education ,1923 Bulletin of the American Academy of Medicine American Academy of Medicine,1911 **Memoirs of the National Academy of Sciences** ,1921 **Bulletin of the American Academy of Medicine** ,1895 **Proceedings of the Academy of Political Science** ,1922 *Bulletin of the New York Academy of Medicine* ,1929 Issues for 1925 37 include the academy s Annual report for 1924 36 **Handbook of the American Academy of Political and Social Science** American Academy of Political and Social Science,1891

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web if you start with 14.82 g of Ca(OH)2 and 16.35 g of H2SO4 determine the limiting reagent b determine the number of moles of H2O produced c determine the number of grams of CaSO4 produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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