

8. Given the following data for a fluid, what is its state at 40°C and 3 kPa?

Saturated property table

$T_s, ^\circ\text{C}$	P_s, kPa	$v_{f,s}, \text{m}^3/\text{kg}$	$v_{g,s}, \text{m}^3/\text{kg}$	$h_{f,s}, \text{kJ/kg}$	$h_{g,s}, \text{kJ/kg}$	$s_{f,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$	$s_{g,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$
40	7.38	.001008	19.52	167.57	2574.3	.5725	8.257
80	47.39	.001029	3.407	334.9	2643.7	1.1343	7.5645
120	198.5	.001060	.8919	503.71	2706.3	1.5276	7.1296

- (A) Saturated liquid
- (B) Superheated vapor
- (C) Compressed liquid
- (D) Saturated vapor

9. Using the same data of Problem 8, what is the fluid's entropy in kJ/(kgK) at 120°C and 80% quality?

- (A) 1.53
- (B) 6.009
- (C) 7.13
- (D) 28.8

10. A quantity of 55 m³ of water passes through a heat exchanger and absorbs 2,800,000 kJ. The exit temperature is 95°C. The entrance water temperature in °C is nearest to:

- (A) 49
- (B) 56
- (C) 68
- (D) 83

11. In any non-quasistatic thermodynamic process, the overall entropy of an isolated system will:

- (A) Increase and then decrease
- (B) Decrease and then increase
- (C) Stay the same
- (D) Increase only

12. The temperature-entropy diagram in the figure represents a:

- (A) Rankine cycle with superheated vapor
- (B) Carnot cycle
- (C) Diesel cycle
- (D) Refrigeration cycle



13. Ethane is burned in 150% theoretical air. Assume complete combustion and that the process takes place in atmospheric air. The air/fuel ratio in 150% theoretical air is most nearly:

- (A) 32 kg air/kg fuel
- (B) 28 kg air/kg fuel
- (C) 24 kg air/kg fuel
- (D) 16 kg air/kg fuel

14. There are 3 kg of air in a rigid container at 250 kPa and 50°C. The gas constant for air is 0.287 kJ/(kgK). The volume of the container, in m³, is nearest to:

- (A) 2.2
- (B) 1.1
- (C) 2.8
- (D) 3.1

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yang sudah kita hitung sebelumnya dengan kedua rumus tersebut diatas untuk mengisi tabel daftar harga jual kita cukup mengetikan nama barang serta harga belinya dan nilai profit serta harga jual akan otomatis muncul

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