



EARTH SCIENCE

INTRODUCTION

COMPONENTS OF STUDY

- Geology: Earth physical & historical
- Metaphysiology: Weather & Climate
- Oceanography: The oceans
- Astronomy: What is external to the atmosphere

THE FOUR SPHERES (See The Water Cycle)

- Lithosphere: The solid earth
- Hydrosphere: The liquid earth
- Atmosphere: The gaseous earth
- Biosphere: Where life exists

CONCERN IN OUR PHYSICAL ENVIRONMENT

- Resources: Water, soil, metallic & non-metallic elements, sunlight, sunlight, wind, etc.
- The capita use of metallic and non-metallic elements is 11 tons per year
- Renewable vs. non-renewable resources
- Environmental and population problems

MINERALS

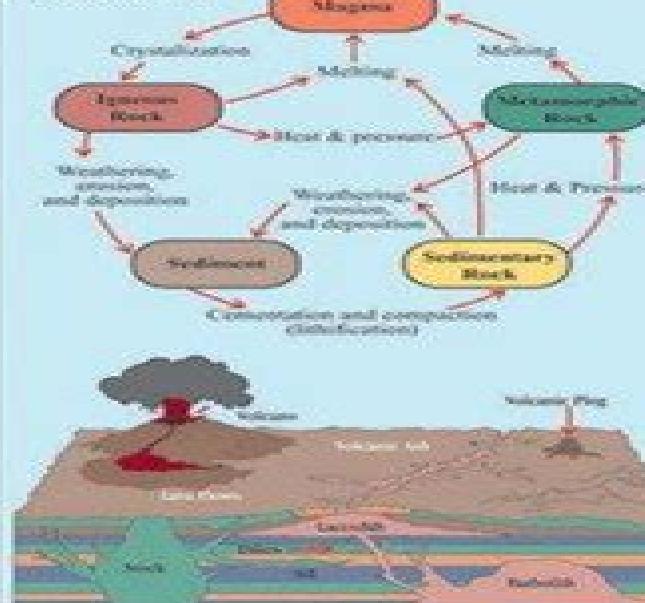
- A mineral is a naturally occurring, inorganic, solid material with a definite atomic structure.
- An ore is a useful mineral resource.
- A. Atom and Minerals**
 - Atoms in ancient Greek means "un-cut" i.e., the Greeks believed the atom to be the smallest particle of matter.
 - Atoms are composed of neutrons, protons and electrons.
 - Atomic number: Number of protons in nucleus.
 - Atomic weight: Average weight of an atom.
 - Isotopes: Atoms with the same atomic number but different numbers of neutrons in the nucleus.
 - Forces: Strong, weak, electromagnetic and gravity.
- Properties of Minerals**
 - Color: Appearance or quality of light from surface.
 - Texture: Nature of light and cause of color.
 - Streak: Color of material in powdered form.
 - HARDNESS: Mohs' Scale (See Figure Below).
 - Scratch Test: Higher numbered materials can scratch lower numbered materials.
 - Shape is determined by cleavage, crystal form and structure.
 - Specific Gravity:
 - The ratio of the weight of a mineral to the weight of an equal volume of water.
 - Density of water = 1 gram/cubic centimeter.
 - Iron, iron = 7.7, aluminum = 2.7, aluminum = 2.2.

MOM'S SCALE		MINERALS
10	Diamond	
9	Corundum	
8	Topaz	
7	Quartz	
6	feldspar	
5	Spatite	
4	Fluorite	
3	Gypsum	
2	Orthoclase	
1	Halite	
0	Table	

C. Mineral Concepts

- Silicates: Minerals with silicon and oxygen.
- Silicates are a large class which forms a pyramidal-shaped structure with oxygen.
- Other forms include simple non-silicate minerals, oxides, sulfides, noble elements, 2-3 element hydroxides and quartz.
- Non-Silicates
- Make up one-fourth of continental crust.
- Carbides: Minerals with carbon and oxygen including calcium, those which are present limestone (calcium) and marble (calcium carbonate).
- Oxides: Oxygen-banded solids, i.e., iron sulfides (FeS), sulfates (MgSO₄), halides (CaCl₂, F).
- Halite: Mineral form of salt.
- Corundum: Plasma, calcium.
- Noble metals: Iron, zinc, gold, silver, nickel.

THE ROCK CYCLE



ROCK TYPES

METAMORPHIC ROCKS

A. Formation

- Metamorphism**: To change form.
- Regional metamorphism**: Material under intense pressure, high temperatures.
- Contact metamorphism**: Material closest to magma.
- Low grade metamorphism**: Shallow metamorphic rocks.
- High grade metamorphism**: Deeper metamorphic rocks.
- foliation**: During metamorphism, parallel mineral lines are formed that are perpendicular to the metamorphic pressure.
- Metamorphic structures**: Features of material that formed under heat with one another.
- Agents of Metamorphism**

B. Heat

- most important agent
- possible energy for chemical reactions
- For example, clay becomes shale at increased or great temperature

C. Chemical Activity

- Metamorphic Rocks**
 - Metamorphic**: Fine grained, dark, shiny rock with randomly arranged minerals.
 - Shale**: Fine grained, often grey, folded rock, light grey along cleavage, pieces of rock broken by pressure.
 - Marble**: Calcium or magnesium-rich rock, formed from limestone.
 - Phyllite**: Shiny, folded rock, more correctly foliated than shale.
 - Schist**: Folded rock, more coarsely grained and w/ higher metamorphic grade than phyllite.
 - Gneiss**: Folded, banded rock w/ the highest metamorphic grade.
 - Quartzite**: Very hard, granular quartz rock, formed from sandstone.

ROCKS IN REACTION SERIES

- is a mineral, which has already formed, reacts w/ other minerals, it will react w/ the remaining magma to produce the next mineral in the sequence. For example, olivine reacts w/ silica, which reacts w/ remaining magma to form pyroxene.

D. High-Mg Shale and High-Silica Series

- Calcium-rich rocks w/ other minerals form small crystals w/ foliation.
- These react w/ sodium in the magma to become mica and more sodium-rich.

E. End of Cooling

- Without everything is altered over, remaining magma w/ have high silica content, and quartz w/ form.
- Without cooling w/ convection, everything that reacted at the same time will remain by the place for ever, another pyroxene, mica and quartz cloud never ever another for major growth.



Earth Science High School Essential Question

RJ Alexander

Earth Science High School Essential Question:

Teaching Discipline-Specific Literacies in Grades 6-12 Vicky I. Zygouris-Coe, 2014-10-30 Comprehensive timely and relevant this text offers an approach to discipline specific literacy instruction that is aligned with the Common Core State Standards and the needs of teachers students and secondary schools across the nation It is essential that teachers know how to provide instruction that both develops content and literacy knowledge and skills and aims at reducing student achievement gaps Building on the research supported premise that discipline specific reading instruction is key to achieving these goals this text provides practical guidance and strategies for prospective and practicing content area teachers and other educators on how to prepare all students to succeed in college and the workforce Pedagogical features in each chapter engage readers in digging deeper and in applying the ideas and strategies presented in their own contexts Classroom Life real 6 12 classroom scenarios and interviews with content area teachers Common Core State Standards Connections College Career and Workforce Connections Applying Discipline Specific Literacies Think Like an Expert habits of thinking and learning specific to each discipline Digital Literacies Differentiating Instruction Reflect and Apply Questions Extending Learning Activities The Companion Website includes Lesson plan resources Annotated links to video files Annotated links to additional resources and information Glossary Flashcards For Instructors All images and figures used in the text provided in an easily downloadable format For Instructors PowerPoint lecture slides The Mastery Learning Handbook Jonathan Bergmann, 2022-10-14 Mastery learning is an instructional approach that empowers every student to progress with confidence Using flexible pacing and targeted supports teachers guide students through a cyclic process of preparation demonstration and formative feedback until there is a mutual agreement between teacher and student that the student is ready to advance In this book educator Jonathan Bergmann a pioneer of the flipped classroom movement walks you step by step through the mastery learning cycle explaining what it entails and providing the templates models and rubrics you need to start using it in your own classroom You ll learn how to Set meaningful measurable and transferable learning objectives that target essential knowledge skills and dispositions Develop a mastery rubric to check for student levels of proficiency Create an assessment plan that ensures positive learning experiences for all Plan and deliver units that incorporate both time shifted direct instruction and collaborative application activities within the classroom space Provide timely differentiated support based on students individual learning needs Informed by trial and error in his own classroom and by discussions with other expert practitioners Bergmann shares commonsense solutions to the major challenges of mastery learning implementation everything from how to manage pacing to how to create multiple versions of tests determine grades and get stakeholder buy in The Mastery Learning Handbook is both an introduction to this exciting instructional approach and a practical resource that K 12 teachers can turn to again and again See for yourself just how effective enjoyable and transformative mastery learning can be Resources in Education , 1992 Serves as an index to Eric reports microform

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Science Curriculum Resource Handbook ,1992 **Intersections of Formal and Informal Science** Lucy Avraamidou,Wolff-Michael Roth,2016-03-10 Science learning that takes place between and at the intersections of formal and informal science environments has not been systematically reviewed to offer a comprehensive understanding of the existing knowledge base Bringing together theory and research this volume describes the various ways in which learning science in various settings has been conceptualized as well as empirical evidence to illustrate how science learning in these settings can be supported

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Table of Contents Earth Science High School Essential Question

1. Understanding the eBook Earth Science High School Essential Question
 - The Rise of Digital Reading Earth Science High School Essential Question
 - Advantages of eBooks Over Traditional Books
2. Identifying Earth Science High School Essential Question
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Earth Science High School Essential Question
 - User-Friendly Interface
4. Exploring eBook Recommendations from Earth Science High School Essential Question
 - Personalized Recommendations
 - Earth Science High School Essential Question User Reviews and Ratings
 - Earth Science High School Essential Question and Bestseller Lists
5. Accessing Earth Science High School Essential Question Free and Paid eBooks
 - Earth Science High School Essential Question Public Domain eBooks
 - Earth Science High School Essential Question eBook Subscription Services
 - Earth Science High School Essential Question Budget-Friendly Options
6. Navigating Earth Science High School Essential Question eBook Formats
 - ePUB, PDF, MOBI, and More
 - Earth Science High School Essential Question Compatibility with Devices
 - Earth Science High School Essential Question Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Earth Science High School Essential Question
 - Highlighting and Note-Taking Earth Science High School Essential Question
 - Interactive Elements Earth Science High School Essential Question
8. Staying Engaged with Earth Science High School Essential Question

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Earth Science High School Essential Question

9. Balancing eBooks and Physical Books Earth Science High School Essential Question

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Earth Science High School Essential Question

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Earth Science High School Essential Question

- Setting Reading Goals Earth Science High School Essential Question
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Earth Science High School Essential Question

- Fact-Checking eBook Content of Earth Science High School Essential Question
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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