

# **Sale and Purchase Agreement**

Sample

THIS AGREEMENT is made the day and year set out in Section 1 of the First Schedule hereto BETWEEN :

**The Vendor** as described in Section 2 of the First Schedule herein; and  
**The Purchaser** as described in Section 2 of the First Schedule herein.

The Vendor and the Purchaser are represented by their respective solicitors as stated in Section 2 of the First Schedule herein.

## **SALE AND PURCHASE**

### **2.1 AGREEMENT TO SELL AND PURCHASE**

The Vendor agrees to sell and the Purchaser agrees to purchase the Property as described in Section 1 of the Second Schedule herein at the Purchase Price as described in Section 3 of the First Schedule herein upon the terms and conditions herein contained. The Vendor agrees not to deal or part with the Property after the execution of this agreement. The completion of the sale shall take place at the Vendor's solicitors' office or the Purchaser's solicitors' office if the Vendor is unrepresented.

### **2.2 MANNER OF PAYMENT**

The time and manner of payment of the Purchase Price shall be as agreed in Section 3 of the First Schedule herein by banker's cheque or bank draft or in cash, which shall be paid towards redemption sum, if any, to the chargee or body or person concerned with such encumbrance and any balance thereof shall be paid to the Vendor. Any shortfall in the redemption sum shall be paid by the Vendor within 14 days from the Purchaser's solicitors' notification.

### **2.3 CONDITIONS OF TITLE**

2.3.1 The sale of the Property is on condition that the title of the Property shall be good and registrable, being free from encumbrances, trusts, caveat, prohibitory order and any other liability capable of being attached to the Property but otherwise subject to all conditions and restrictions in the title.

2.3.2 If the title is subject to a restriction of title, the Vendor shall at his own cost, within six(6) months from the date of this Agreement, obtain the necessary consent from the relevant authority to the sale and transfer herein, failing which the Purchaser shall be entitled to terminate this Agreement and obtain a full refund of whatsoever Purchase Price monies paid without interest.

### **2.4 CONDITIONS FOR SUBDIVIDED BUILDING WITHOUT TITLE**

2.4.1 The Vendor has performed and complied with all terms and conditions in the Principal Agreement, by-laws and other related covenants and the Property is not subject to any foreclosure proceedings by any financier and or parties.

# Accounting Practice Sales Agreement

**California (State).**

**Accounting Practice Sales Agreement:**

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California. Court of Appeal (2nd Appellate District). Records and Briefs California (State).,    **Tax Court Digest** United States. Tax Court,1951    National Painters Magazine,1926    The Grand Rapids Furniture Record,1924    Accountants' Index American Institute of Certified Public Accountants,1928    **The Painters Magazine and Paint and Wall Paper Dealer**,1926    **Reports of the Tax Court of the United States** United States. Tax Court,1930    New York Law of Contracts Briscoe Baldwin Clark,1922    N.A.C.A. Bulletin,1927    **NACA Bulletin** National Association of Cost Accountants (U.S.),1926

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3 655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 000731 moles making the experimental molar mass

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calculate  $k_{a1}$  and  $k_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1  
hcl aq naoh aq nacl aq h<sub>2</sub>o 1 hcl a q naoh a q nacl a q h 2 o l one mole of hcl hcl would be fully neutralized by one mole of  
naoh naoh

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standardization of sodium hydroxide naoh and oxalic acid h<sub>2</sub>c<sub>2</sub>o<sub>4</sub> 2h<sub>2</sub> o and the molar mass of an unknown solid acid using  
the molar mass of the standardized sodium hydroxide solution

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a ph titration is used to determine the concentration of an acid or a base titrations play an important role in determining  
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acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a  
solution of known strength is added to a certain volume of a treated sample containing an

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reaction with a 2nd substance h<sub>2</sub>c<sub>2</sub>o aq 2naoh aq na<sub>2</sub>c<sub>2</sub>o<sub>4</sub> 4h<sub>2</sub>o to differentiate an acid from a base in a solution a ph scale is  
used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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