



Chapter 14 - DNA: The Genetic Material

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1. What won the Nobel prize for DNA structure and what?	Crick, Watson and Wilkins in 1962
2. What are the components of a nucleotide?	sugar, phosphate group, nitrogenous base
3. What is the difference between deoxyribose and ribose?	deoxyribose has H attached to its 2' C while ribose has OH attached to its 2' C
4. Which nitrogenous bases are purines?	Adenine and Guanine
5. Which nitrogenous bases are pyrimidines?	Cytosine, thymine, uracil
6. Are purines single or double ringed?	double
7. Are pyrimidines single or double ringed?	single
8. What is a phosphodiester bond?	one that forms between the 3' phosphate group of one sugar and 5' hydroxyl group of adjacent sugar



Biology The Genetic Material Study Guide Answers

**William K. Purves, Edward
Dzialowski, Lindsay Goodloe, Betty
McGuire, Nancy Guild, Paula Mabee**

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Life: The Science of Biology Study Guide William K. Purves, Edward Dzialowski, Lindsay Goodloe, Betty McGuire, Nancy Guild, Paula Mabee, 2003-12-26 New edition of a text presenting underlying concepts and showing their relevance to medical agricultural and environmental issues Seven chapters discuss the cell information and heredity evolutionary process the evolution of diversity the biology of flowering plants and of animals and ecology and biogeography Topics are linked by themes such as evolution the experimental foundations of knowledge the flow of energy in the living world the application and influence of molecular techniques and human health considerations Includes a CD ROM which covers some of the subject matter and introduces and illustrates 1 700 plus key terms and concepts Annotation copyrighted by Book News Inc Portland OR Biology, Study Guide Gilbert D. Brum, Larry McKane, Gerald Karp, 1993-10-28 This lively richly illustrated text makes biology relevant and appealing revealing it as a dynamic process of exploration and discovery Portrays biologists as they really are human beings with motivations misfortunes and mishaps much like everyone has Encourages students to think critically solve problems apply biological principles to everyday life **Study Guide for The Developing Person Through Childhood and Adolescence 6e** Richard O. Straub, Kathleen Stassen Berger, 2002-11-07 This title is the study guide which accompanies The Developing Person Through Childhood and Adolescence 6th edition ISBN 0716752573 **Study Guide to Accompany Biology by Karen Arms and Pamela S. Camp** Russell C. Hollingsworth, 1979 **Life Study Guide** David E. Sadava, Gordon H. Orians, Craig Heller, William K. Purves, 2006-12-22 Especially helpful for AP Biology students each chapter of the study guide offers a variety of study and review tools The contents of each chapter are broken down into both a detailed review of the Important Concepts covered and a boiled down Big Picture snapshot The guide also covers study strategies common problem areas and provides a set of study questions both multiple choice and short answer Student Study Guide Liebaert, 2002-08 by Richard Liebaert Linn Benton Community College Students can master key concepts and earn a better grade with the thought provoking exercises found in this study guide A wide range of questions and activities help students test their understanding of biology The Student Study Guide also includes references to student media activities on the Campbell Biology CD ROM and Web Site Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Denniston Katherine, 2006 **Study Guide to Accompany Biology: A Journey into Life** Russell V. Skavaril, 1991 Study Guide and Solutions Manual Bruce A. Chase, Peter J. Russell, 2005-06 This student resource contains chapter outlines of text material solutions to all end of chapter problems key terms suggestions for analytical approaches problem solving strategies and a variety of additional questions for student practice Also featured are questions that relate to chapter specific animations and iActivities **Study Guide and Solutions Manual for Students, to Accompany General Genetics** David Freifelder, Leon A. Snyder, 1985 Study Guide: Sg Concepts in Biology Eldon D.. Enger, Enger, Frederick C.. Ross, 2003

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web correct answer c both a and b q2 the density of 1 m solution of hcl is 1 0585 g ml the molality of the solution is 1 0585 1 0 10 0 0585 correct answer b 1 explanation molality begin array l m left frac 1000 m 1000d mm right end array

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web sep 13 2023 the only difference between molality and molarity is in the denominator molality is measured in kilograms of solvent whereas molarity is measured in liters of solution when studying the properties of solutions related to vapour

pressure and temperature changes concentrations expressed in molality are used

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web 1. molarity $M = \frac{\text{mass}}{\text{molar mass} \times \text{volume}}$ $\frac{0.4500 \text{ L} \times 0.825 \text{ g/L}}{141.9579 \text{ g/mol} \times 0.0129 \text{ m}^3} = 2.62 \text{ molality}$ $\frac{0.825 \text{ g/L} \times 141.9579 \text{ g/mol}}{0.00581158 \text{ mol} \times 0.00581158 \text{ mol} \times 4500 \text{ kg}} = 0.0129 \text{ m}^3$ mole fraction $\frac{2 \text{ hpo}}{4 \times 0.825 \text{ g/L} \times 141.9579 \text{ g/mol} \times 0.00581158 \text{ mol}} = 2.0$ $\frac{450 \text{ g}}{0.18 \text{ L}} = 0.15 \text{ g/mol}$ $\frac{24 \times 97918401 \text{ mol}}{1}$

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web apr 7 2014 molarity moles of solute liters of solution $\frac{8.4 \text{ L} \times 2 \text{ a}}{1} = 16.8$ a first convert 250 ml to liters $\frac{250}{1000} = 0.25$ then calculate molarity $\frac{5 \text{ moles}}{0.25 \text{ liters}} = 20 \text{ M}$ 3. c a solution with molarity 2 requires 2 M of NaOH per liter so $4 \times 2 = 8 \text{ M}$ 4. a a solution of molarity 1.5 M requires 1.5 mol of NaOH to every litre of solvent

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web answer the correct option is c explanation the molarity of a mixture M_{mix} can be calculated using the following formula $M_{\text{mix}} = \frac{M_1 V_1 + M_2 V_2 + \dots + M_n V_n}{V_1 + V_2 + \dots + V_n}$ $\frac{1 \times 480 + 2 \times 520 + 480 + 520 + 720 + 624 + 1000 + 1344}{1000} = 1.344 \text{ M}$ q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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web 1. calculate the molarity of a solution which contains 0.40 mol of a substance dissolved in 1.6 L of a solution 2. what is the molarity of a solution containing 325 g of NaCl dissolved in 750 ml of solution 3. 140 g of KCl is dissolved in 600 ml of water what is the molarity 4. 724.4 g of ammonium phosphate in 4500 ml of alcohol

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web determine the molarity for each of the following solutions 0.444 mol of COCl_2 in 0.654 L of solution 98.0 g of phosphoric acid H_3PO_4 in 1.00 L of solution 0.2074 g of calcium hydroxide Ca(OH)_2 in 40.00 ml of solution 10.5 kg

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