



EARTH SCIENCE

INTRODUCTION

COMPONENTS OF STUDY

- Geology: Earth physical & historical
- Metaphysiology: Weather & Climate
- Oceanography: The oceans
- Astronomy: What is external to the atmosphere

THE FOUR SPHERES (See The Water Cycle)

- Lithosphere: The solid earth
- Hydrosphere: The liquid earth
- Atmosphere: The gaseous earth
- Biosphere: Where life exists

CONCERN IN OUR PHYSICAL ENVIRONMENT

- Resources: Water, soil, metallic & non-metallic elements, sunlight, sunlight, wind, etc.
- The capita use of metallic and non-metallic elements is 11 tons per year
- Renewable vs. non-renewable resources
- Environmental and population problems

MINERALS

- A mineral is a naturally occurring, inorganic, solid material with a definite atomic structure.
- An ore is a useful mineral resource.

A. Atom and Minerals

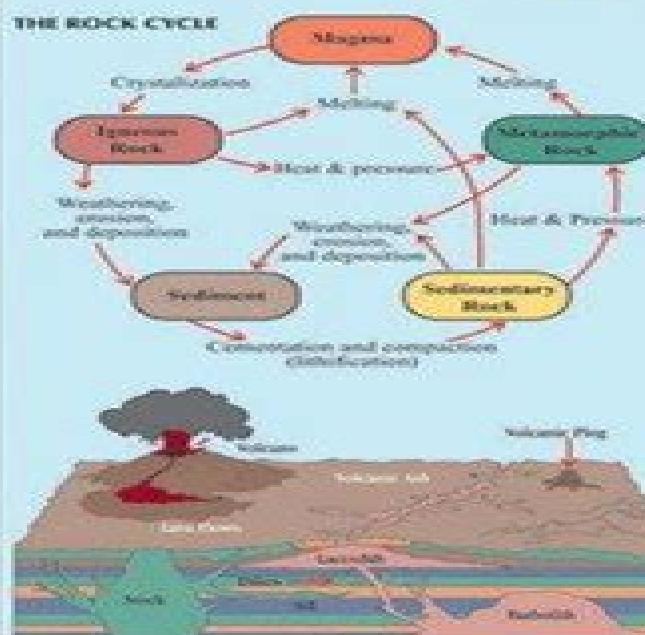
- Atoms in ancient Greek means "un-cut" i.e., the Greeks believed the atom to be the smallest particle of matter.
- Atoms are composed of neutrons, protons and electrons.
- Atomic number: Number of protons in nucleus.
- Atomic weight: Average weight of an atom.
- Isotopes: Atoms with the identical atomic number, but different numbers of neutrons in the nucleus.
- Forces: Strong, weak, electromagnetic and gravity.
- Properties of Minerals**
 - Color: Appearance or quality of light from surface.
 - Texture: Nature of light and cause of color.
 - Streak: Color of material in powdered form.
 - HARDNESS: Mohs' Scale (See Figure Below).
 - Scratch Test: Higher numbered materials can scratch lower numbered materials.
 - Shape is determined by cleavage, crystal form and structure.
- Specific Gravity: The ratio of the weight of a mineral to the weight of an equal volume of water.
- Density of water = 1 gram/cubic centimeter.
1 g/cm³ = 7.7, aluminum = 2.7, iron = 7.2.

| MOM'S SCALE | | MINERALS |
|-------------|----------|----------|
| 10 | Diamond | |
| 9 | Corundum | |
| 8 | Topaz | |
| 7 | Quartz | |
| 6 | feldspar | |
| 5 | Spatite | |
| 4 | Fluorite | |
| 3 | Garnet | |
| 2 | Pyrope | |
| 1 | Redite | |
| 0 | Table | |

MINERAL CONCEPTS

- Silicates: Minerals with silicon and oxygen.
- Silicates are a large class which forms a pyramidal-shaped structure with oxygen.
- Other forms include simple non-silicate minerals, oxides, sulfides, noble elements, 2-3 element hydroxides and quartz.
- Non-Silicates
- Make up one-fourth of continental crust.
- Carbides: Minerals with carbon and oxygen including calcium, those which are present limestone (calcium) and marble (calcium carbonate).
- Oxides: Oxygen-banded solids, i.e., iron sulfides (FeS), sulfates (MgSO₄), halides (CaCl₂, F).
- Halite: Mineral form of salt.
- Corundum: Plasma, calcium.
- Noble metals: Iron, zinc, gold, silver, nickel.

THE ROCK CYCLE



IGNEOUS ROCKS

A. Formation

- Extrusive igneous rocks are formed in volcanic eruptions at surface.
- Extrusive or plutonic igneous rocks are produced within the earth.
- Magma**
 - Is a mixture mainly of silicate minerals.
 - Has gases, such as water vapor.
 - Temperature there are about 2000 °K.
 - Causes all of cooling, chemical composition, and amount of gases.
 - Produces intrusive igneous rocks.
- Lava**
 - Similar to magma.
 - Most gaseous elements have escaped.
 - Produces extrusive (volcanic) igneous rocks.
- Metamorphic**
 - Can stabilize different patterns of ions.
 - The rate of cooling strongly affects crystal size.
 - Slow cooling = Large crystals.
 - Quick cooling = Small crystals.
- Amorphous (non-crystalline) pattern of ions**
 - rocks with instant cooling (quenching).
 - produces rocks referred to as glass.
- Classifying Igneous Rocks**
 - By texture: size and arrangement of interlocking crystals.
 - Fast-cooled
 - Form at surface or within upper crust.
 - Crystals left the gas bubbles are called vesicles.
 - Slow-cooled
 - Formed deep within the interior.
 - Large crystal embedded in matrix of smaller crystals.
 - By mineral composition
 - Depends on composition of magma and cooling.
 - Some magma can create rocks of varying mineral content.
 - An igneous rock contains minerals crystallized first (or very early) to form.

ROCK TYPES

METAMORPHIC ROCKS

A. Formation

- Metamorphism: To change form.
- Regional metamorphism: Material under intense pressure, high temperature.
- Contact metamorphism: Metamorphism caused by proximity to magma.
- Low grade metamorphism: Shallow metamorphic rock.
- High grade metamorphism: Deeper metamorphic rock.
- foliation: During metamorphism, minerals are arranged in lines with one another.
- Agents of Metamorphism**
- Heat**
 - most important agent.
 - possible energy for chemical reactions.
 - For example, clay converts into a mineral at great temperature.
- Pressure**
- Chemical Activity**: Most common chemical agent in nature.

Types of Metamorphic Rocks

- Metamorphic**: Fine grained, dark, shiny rock with randomly arranged minerals.
- Schist**: Fine grained, often grey, foliated rock, right below along cleavage, pieces of mica flakes aligned by pressure.
- Migmatite**: Contains one migmatized rock, formed from foliation.
- Phyllite**: Shiny, foliated rock more correctly foliated than slate.
- Slate**: Foliated rock, more coarsely grained and wavy foliated metamorphic grade than phyllite.
- Gneiss**: Foliated, banded rock of the middle metamorphic grade.
- Quartzite**: Very hard, granular quartz rock, formed from sandstone.

ROCKS' REACTION SERIES

- is a mineral, which has already formed, reacts with other minerals, it will react with the remaining magma to produce the next mineral in the sequence. For example, calcium-rich feldspar reacts with remaining magma to form plagioclase.

B. Plagioclase and other Olivine Series

- Calcium-rich rocks will react with remaining magma to form plagioclase.
- Then react with sodium in the magma to form feldspar and more sodium-rich.

C. End of Cooling

- When everything is altered and remaining magma with have high sodium content, and quartz will form.
- Without cooling is complete, everything that reacted at the same time will remain by the glass. In this process, pyroxene and olivine will react over and over again for further growth.



Earth Science High School Essential Question

M Mark

Earth Science High School Essential Question:

Teaching Discipline-Specific Literacies in Grades 6-12 Vicky I. Zygouris-Coe, 2014-10-30 Comprehensive timely and relevant this text offers an approach to discipline specific literacy instruction that is aligned with the Common Core State Standards and the needs of teachers students and secondary schools across the nation It is essential that teachers know how to provide instruction that both develops content and literacy knowledge and skills and aims at reducing student achievement gaps Building on the research supported premise that discipline specific reading instruction is key to achieving these goals this text provides practical guidance and strategies for prospective and practicing content area teachers and other educators on how to prepare all students to succeed in college and the workforce Pedagogical features in each chapter engage readers in digging deeper and in applying the ideas and strategies presented in their own contexts Classroom Life real 6 12 classroom scenarios and interviews with content area teachers Common Core State Standards Connections College Career and Workforce Connections Applying Discipline Specific Literacies Think Like an Expert habits of thinking and learning specific to each discipline Digital Literacies Differentiating Instruction Reflect and Apply Questions Extending Learning Activities The Companion Website includes Lesson plan resources Annotated links to video files Annotated links to additional resources and information Glossary Flashcards For Instructors All images and figures used in the text provided in an easily downloadable format For Instructors PowerPoint lecture slides *The Mastery Learning Handbook* Jonathan Bergmann, 2022-10-14 Mastery learning is an instructional approach that empowers every student to progress with confidence Using flexible pacing and targeted supports teachers guide students through a cyclic process of preparation demonstration and formative feedback until there is a mutual agreement between teacher and student that the student is ready to advance In this book educator Jonathan Bergmann a pioneer of the flipped classroom movement walks you step by step through the mastery learning cycle explaining what it entails and providing the templates models and rubrics you need to start using it in your own classroom You ll learn how to Set meaningful measurable and transferable learning objectives that target essential knowledge skills and dispositions Develop a mastery rubric to check for student levels of proficiency Create an assessment plan that ensures positive learning experiences for all Plan and deliver units that incorporate both time shifted direct instruction and collaborative application activities within the classroom space Provide timely differentiated support based on students individual learning needs Informed by trial and error in his own classroom and by discussions with other expert practitioners Bergmann shares commonsense solutions to the major challenges of mastery learning implementation everything from how to manage pacing to how to create multiple versions of tests determine grades and get stakeholder buy in The Mastery Learning Handbook is both an introduction to this exciting instructional approach and a practical resource that K 12 teachers can turn to again and again See for yourself just how effective enjoyable and transformative mastery learning can be *Resources in Education* , 1992 Serves as an index to Eric reports microform

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Science Curriculum Resource Handbook ,1992 **Intersections of Formal and Informal Science** Lucy Avraamidou,Wolff-Michael Roth,2016-03-10 Science learning that takes place between and at the intersections of formal and informal science environments has not been systematically reviewed to offer a comprehensive understanding of the existing knowledge base Bringing together theory and research this volume describes the various ways in which learning science in various settings has been conceptualized as well as empirical evidence to illustrate how science learning in these settings can be supported

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Publisher **Proceedings of the ... Annual Conference** Science Teachers Association of New York State, 1904

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