

ACID & BASE

- Provide optimal cell function
 - Cells secrete **acid** and bases to maintain proper pH for enzymes to work.
- They both provide homeostatic regulation of pH of the body's ECF (extracellular fluid)
 - The Carbonic Acid-Bicarb Buffer System = pH homeostasis of blood

ACID

CO₂ is regulated by the lungs + contributes to acid

- found in the stomach + urine + kidneys *
 - ex: acid reflux, BPH, kidney failure

BASE

Bicarb HCO_3 = form of carbon dioxide in our blood

- found in intestines *
 - Regulated, excreted + reabsorbed in kidneys *
 - Ex: C-diff → bacteria in colon

2 FUNCTIONS

Metabolic + Respiratory

METABOLIC

primary change is the concentration of bicarbonate.

METABOLIC ACIDOSIS

bicarb ↓ r/t the amount of acid in the body

- Increased Hydrogen ions
 - CAUSES: kidney failure, DKA, lactic acidosis (shock, sepsis), severe diarrhea
 - put pt. on bicarb drip to increase levels
- Respiratory compensation:
 - Increased RR to rid of acid

METABOLIC ALKALOSIS

excess bicarb r/t the amount of acid in the body

- Decreased Hydrogen ions
 - CAUSES: loss of gastric fluids, antacid overdose
 - ex: too much NGT, suctioning, hypokalemia, vomiting

RESPIRATORY

primary change is the concentration of carbonic acid

RESPIRATORY ACIDOSIS

CO₂ is retained + inadequate O₂ exchange → increases amount of carbonic acid in the body

- CAUSES: Hypoventilation
- ex: Asthma, COPD, Pneumonia
- Decreased RR
 - Sleep apnea, drug overdose, head injuries

Compensation?

- Increasing RR

RESPIRATORY ALKALOSIS

Too much CO₂ is lost + carbonic acid levels fall

- Pt has increased RR
 - ex: hyperventilation due to anxiety attack

Compensation?

- slow + shallow breaths

Acid And Base Study Guide

David R. Klein, Laurie S. Starkey

Acid And Base Study Guide:

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High-yield General Chemistry Stephen D. Bresnick, 1996. A study guide in question and answer format for basic chemistry.

Organic Chemistry Michael B. Smith, 2022-09-23. Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts. **Organic Chemistry An Acid Base Approach** provides a framework for understanding the subject that goes beyond mere memorization. Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry. This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the previous editions. Highlights of the Third Edition include extensively revised chapters that improve the presentation of material. Features the contributions of more than 65 scientists highlighting the diversity in organic chemistry. Features the current work of over 30 organic chemists highlighting the diversity in organic chemistry. Many new reactions are featured that are important in modern organic chemistry. Video lectures are provided in a mov format accessible online as a built in ancillary for the book. Instructor and Student Resources includes scientist images and solutions manual for instructors. The third edition of **Organic Chemistry An Acid Base Approach** constitutes a significant improvement upon a unique introductory technique to organic chemistry. The reactions and mechanisms it covers are the most fundamental.

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