

Inorganic Chemistry ACS: Study Guide Exam|334 Questions with Accurate Answers

Ionization energy - -energy required to remove the least tightly bound electron from a neutral atom in the gas phase

- periodic trend of ionization energy - -highest at top right-smaller electron=harder to remove
- Why is a half filled subshell so stable? - -it serves to maximize the stabilizing interactions while minimizing the destabilizing interactions among electrons
- exchange interaction - -pic, stabilizing, result of electrons pairing in degenerate orbitals with parallel spin
- pairing energy - -destabilizing, coulomb interaction, pic, energy of electron-electron repulsion in a filled orbital
- Is it easier to ionize a high energy or low energy electrons - -high energy electron-already contains more energy so it requires less energy input
- What happens when a 3d series metal is ionized? - -the first electron to be ionized will come from the 4s orbital, the other s electron will enter the d orbital ($4s03dn+1$)
- lanthanide contraction - -reduction in atomic radius following the lanthanide series, contrary to the overall trend observed for the periodic table
- lanthanides - -elements 57-71, first appearance of f orbitals, f orbitals are poor at shielding so any electrons added will have a higher Z_{eff} , shrinking the radius
- Slater's rules - -tell us what the effective nuclear charge will be, $Z_{eff}=Z-\sigma$, Z is the atomic number, σ =sum of the number of electrons in a given shell multiplied by a weighting coefficient (page 1)
- Shielding - -the reduction in charge attraction between the nucleus and electrons due to electrons between the nucleus and the electron in question, it is considered to be between if it has a lower energy

Acs Final Exam Study Guide Inorganic Chemistry

Joshua Rueda

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Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes **The Education Index**, 1991

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