

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|--|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | _____ a base |
| c. Conjugate Acid | _____ | the remaining substance when a proton is lost from an acid |
| d. Conjugate Base | _____ | a substance that produces OH^- in solution |
| e. Bronsted-Lowry Acid | _____ | a substance producing H^+ in solution |
| f. Bronsted-Lowry Base | _____ | a proton acceptor |
| | | a proton donor |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are H_2O conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- , SO_4^{2-}
- HBr , Br^-
- H_3PO_4 , PO_4^{3-}
- H_2O , HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Acids And Bases Guided Answer Key

**T. W. Graham Solomons, Craig B.
Fryhle**



Acids And Bases Guided Answer Key:

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Pocket Guide to Fluid, Electrolyte, and Acid-base Balance Mima M. Horne, 1993 Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson Norman E. Griswold, Henry Fuller Holtzclaw, 1988 The Brass World and Platers Guide, 1920 **Organic Chemistry Study Guide & Problems Book** Brent L. Iverson, 1998 Companion to Organic chemistry 2nd ed by William H Brown Provides a detailed section by section overview of the major points covered in the text. All of the problems from the text are reprinted here with detailed stepwise solutions. *Student's Guide to Fundamentals of Chemistry, Third Edition, Brescia, Arents, Meislich, Turk* Donald Bracken, 1975

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