

### Naming Ionic Compounds Practice Worksheet

Name the following ionic compounds:

- 1)  $\text{NH}_4\text{Cl}$  \_\_\_\_\_
- 2)  $\text{Fe}(\text{OH})_3$  \_\_\_\_\_
- 3)  $\text{TiB}_2$  \_\_\_\_\_
- 4)  $\text{Cu}_3\text{P}$  \_\_\_\_\_
- 5)  $\text{SnSe}_2$  \_\_\_\_\_
- 6)  $\text{GaAs}$  \_\_\_\_\_
- 7)  $\text{Pb}(\text{SO}_4)_2$  \_\_\_\_\_
- 8)  $\text{Ba}(\text{HCO}_3)_2$  \_\_\_\_\_
- 9)  $\text{Mn}_2(\text{SO}_4)_3$  \_\_\_\_\_
- 10)  $\text{Al}(\text{CN})_3$  \_\_\_\_\_

Write the formulas for the following compounds:

- 11) chromium (VI) phosphate \_\_\_\_\_
- 12) vanadium (V) carbonate \_\_\_\_\_
- 13) tin (II) nitrite \_\_\_\_\_
- 14) cobalt (III) oxide \_\_\_\_\_
- 15) titanium (II) acetate \_\_\_\_\_
- 16) vanadium (V) sulfide \_\_\_\_\_
- 17) chromium (III) hydroxide \_\_\_\_\_
- 18) lithium iodide \_\_\_\_\_
- 19) lead (II) nitride \_\_\_\_\_
- 20) silver bromide \_\_\_\_\_

### Lots of Ionic Naming Practice Problems

Name the following ionic compounds:

- 1)  $\text{NaBr}$  \_\_\_\_\_
- 2)  $\text{Sr}(\text{OH})_2$  \_\_\_\_\_
- 3)  $\text{V}_2(\text{SO}_4)_3$  \_\_\_\_\_
- 4)  $\text{NH}_4\text{F}$  \_\_\_\_\_
- 5)  $\text{CaCO}_3$  \_\_\_\_\_
- 6)  $\text{NiPO}_4$  \_\_\_\_\_
- 7)  $\text{Li}_2\text{SO}_4$  \_\_\_\_\_
- 8)  $\text{Zn}_3\text{P}_2$  \_\_\_\_\_
- 9)  $\text{Sr}(\text{C}_2\text{H}_5\text{O}_2)_2$  \_\_\_\_\_
- 10)  $\text{Cu}_2\text{O}$  \_\_\_\_\_
- 11)  $\text{Ag}_3\text{PO}_4$  \_\_\_\_\_
- 12)  $\text{YCO}_3$  \_\_\_\_\_
- 13)  $\text{SnS}_2$  \_\_\_\_\_
- 14)  $\text{Ti}(\text{CN})_3$  \_\_\_\_\_
- 15)  $\text{KMnO}_4$  \_\_\_\_\_
- 16)  $\text{Pb}_2\text{N}_3$  \_\_\_\_\_
- 17)  $\text{CoCO}_3$  \_\_\_\_\_
- 18)  $\text{CdSO}_4$  \_\_\_\_\_
- 19)  $\text{Cu}(\text{NO}_3)_2$  \_\_\_\_\_
- 20)  $\text{Fe}(\text{HCO}_3)_2$  \_\_\_\_\_

# Answers To Ionic Compound Study Guide

**H. Stephen Stoker,Danny V. White**

## Answers To Ionic Compound Study Guide:

*Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

*Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application.

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes. Also includes solutions and explanations to odd numbered exercises.

*Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition* John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter.

**Study Guide and Solutions Manual to Accompany Organic Chemistry, 11th Edition** T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2013-03-25 This is the study guide and solutions manual to accompany Organic Chemistry 11th Edition.

*Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry* Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana. Sue Baker Publishing News

**Study Guide with Answers to Selected Problems** H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test. The answers section includes answers for all odd numbered end of chapter exercises.

**Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Answers To Ionic Compound Study Guide**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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