

GENERAL

- Organic molecules : contains C, forms backbone and branched structures of large diverse molecules as rings or chains, MUST form 4 covalent bonds
 - Hydro-carbons : contains C & H & sometimes O, alkanes (C_nH_{2n+2}), alkenes (C_nH_{2n}), alkynes (C_nH_{2n-2})
-

1.1 THE CHEMISTRY OF LIFE ✓

- Know what intramolecular and intermolecular forces are

Intermolecular Forces: attraction BETWEEN molecules (e.g. hydrogen bonding), like "outer"

Intramolecular Forces: attractions/forces within the molecules (e.g. ionic, polar covalent and covalent bonding)

- Know the difference between covalent and ionic

Ionic: e^- transfer; where resulting ions attract (metal → non-metal)

Covalent: Atoms share e^- to fill valence shell for stable octet (generally 2 non-metals)

- Know what electronegativity is and what electronegativity difference is

Electronegativity: The measure of an atom's attraction to shared electrons

EN Difference: Determines what type of bond there is


- 0.0 — 0.4 : slightly/non-polar covalent (electronically symmetrical, X : X)
- 0.4 — 1.7 : polar covalent (partial charges, $S^+ \& S^-$)
- 1.7 + : ionic bond (full charges, $M^+ X^-$)

- Know the difference between polar and nonpolar

Polarity - partial pos or neg charge @ ends of molecule (one with most EN has partial neg)

Polar: An unequal sharing of electrons; one atom attracts the electrons more than the other does

Non-polar: balanced out, usually gases

- Hydrogen bond - the attractive force between the partially charged H of one molecule & a partially negative charged atom in another molecule
- Water molecule has bent shape due to electrons on O pushing H's down 

- Know what an isotope and radioisotope is

Isotope: # of p^+ same, # of n^0 different, thus the mass # is different

Radioisotope: Radioactive version, decomposes over time

Bio Study Guide Biochemistry

Marcy Osgood, Karen Ocorr



Bio Study Guide Biochemistry:

Biochemistry Basics Milin Kurup, 2020-12-21 The Biochemistry Basics Biochemistry and Molecular Biology Study Guide was created by a renowned student from the University of Florida and includes all notes diagrams and study guides for all the important subjects covered in Biochemistry Molecular Biology Genetics and Microbiology Milin Kurup is a double major in B S Microbiology and Cognitive and Behavioral Neuroscience student from the University of Florida In addition to his degree Milin is a UF Biochemistry BCH4024 Study Instructor Group Leader a Microbiology MCB3020L Teaching Assistant a Genetics PCB4522 Teaching Assistant and a Neuroscience Research Assistant at the University of Florida While many of these classes cover high density material this study guide hopes to organize and condense the whole curriculum into short page review sheets In the author s time of instruction and study he organized a collection of all reactions mechanisms processes and concepts all studied in Biochemistry Genetics and Microbiology Overall this biochemistry study guide covers topics such as biomolecule structures Protein Carbohydrate Nucleic Acids and Lipids biomolecules function biomolecule metabolism Protein Metabolism Carbohydrate Metabolism Nucleic Acid Metabolism and Lipid Metabolism physiological biochemical relationships genetics and biological microbiological biochemical processes Overall the guide is organized into 13 page summaries of each specific topic and acts as a study guide for those who hope to study individual concepts in detail All sections include detailed diagrams color coded notes labeled illustration and detailed descriptions for effective comprehension In addition to class studies many students also have used this study guide as an MCAT review guide The short and condensed review pages have helped many student organize and categorize important topics as they continue to study for the MCAT Ultimately this organized set can be extremely useful for students review especially before class exams school projects standardized test and much more

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