

# ACIDS, BASES AND SALTS

<b>Definition of an acid</b> An acid is a compound which when dissolved in water produces hydrogen ions as the only positively charged ions	<b>What causes the acidic properties of acids?</b> The hydrogen ions ( $H^+$ ) cause the acidic properties and these are formed in the presence of water.
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<b>Another term to refer to an acid</b> An acid is called a proton donor	<b>Why an acid is also called a proton donor?</b> It's because an acid provides protons or hydrogen ions ( $H^+$ ) to other substances during the reaction.
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<b>Substance to which an acid provides protons</b> BASE	<b>A base is then called</b> PROTON ACCEPTOR	<b>Why the base is called a proton acceptor?</b> Because the base accepts hydrogen ions from acids	<b>Equation for the reaction between acid and base</b> $H^{+}_{(aq)} + OH^{-}_{(aq)} \longrightarrow H_2O_{(l)}$
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<b>Common laboratory acids</b>		<b>These three common laboratory acids are also called</b> MINERAL ACIDS	<b>Why these acids are also called mineral acids?</b> They are derived from mineral salts ie chlorides for $HCl$ , sulphates for $H_2SO_4$ and nitrates for $HNO_3$
Hydrochloric acid	$HCl$		
Sulphuric acid	$H_2SO_4$		
Nitric acid	$HNO_3$		

<b>Other mineral acids known</b>	<b>Mineral salts from which the acid is derived</b>	<b>Organic acids known</b>	<b>Naturally occurring acids</b>
Sulphurous acid	$H_2SO_3$	Derived from sulphites	CITRIC ACID from lemons
Carbonic acid	$H_2CO_3$	Derived from carbonates	TARTARIC ACID from grapes
Phosphoric acid	$H_3PO_4$	Derived from phosphates	ACETIC ACID from vinegar
Nitrous acid	$HNO_2$	Derived from nitrates	LACTIC ACID from sour milk
		Methanoic acid ( $HCOOH$ )	Hydrochloric acid from digestive juices

<b>Whenever an acid is dissolved in water, it produces</b> HYDROGEN IONS	<b>Term given to the number of hydrogen ions produced by one molecule of an acid</b> BASICITY OF AN ACID	<b>Definition of basicity of an acid</b> BASICITY of an acid is the <u>number of hydrogen ions produced by one molecule of an acid in aqueous solution</u> .
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# Acids Bases And Salts Guide

**Susan McMurry**

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practitioners from around the world to discuss and exchange ideas on the latest trends and advancements in modern education and information management. The conference program featured a diverse range of research topics including educational technology, digital learning, information systems, and knowledge management. With a focus on exploring innovative approaches and strategies, the conference provided a platform for participants to present their research findings and share insights on the future development of the field. Distinguished speakers included Prof. Qing Ding from Huazhong University of Science and Technology, China; Prof. Longkai Wu from Central China Normal University, China; Assoc. Prof. Lim Chee Leong from Taylor's University, Malaysia; and Assoc. Prof. Teh Sin Yin from Universiti Sains Malaysia, Malaysia. These experts delivered keynote speeches offering valuable perspectives and stimulating discussions on the conference themes. The 4th International Conference on Modern Education and Information Management (ICMEIM 2023) played a significant role in shaping the future development of the field. It provided a platform for researchers and practitioners to share their knowledge, explore emerging trends, and address key challenges in modern education and information management. By facilitating collaboration and promoting interdisciplinary dialogue, the conference contributed to the advancement of innovative practices and strategies in this rapidly evolving field. We extend our sincere appreciation to all participants, presenters, organizers, and sponsors for their valuable contributions in making the ICMEIM a success. We look forward to future editions of the conference and the continued growth and advancement of the field.

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## **Table of Contents Acids Bases And Salts Guide**

1. Understanding the eBook Acids Bases And Salts Guide
  - The Rise of Digital Reading Acids Bases And Salts Guide
  - Advantages of eBooks Over Traditional Books
2. Identifying Acids Bases And Salts Guide
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Acids Bases And Salts Guide
  - User-Friendly Interface
4. Exploring eBook Recommendations from Acids Bases And Salts Guide
  - Personalized Recommendations
  - Acids Bases And Salts Guide User Reviews and Ratings
  - Acids Bases And Salts Guide and Bestseller Lists
5. Accessing Acids Bases And Salts Guide Free and Paid eBooks
  - Acids Bases And Salts Guide Public Domain eBooks
  - Acids Bases And Salts Guide eBook Subscription Services
  - Acids Bases And Salts Guide Budget-Friendly Options

### 6. Navigating Acids Bases And Salts Guide eBook Formats

- ePUB, PDF, MOBI, and More
- Acids Bases And Salts Guide Compatibility with Devices
- Acids Bases And Salts Guide Enhanced eBook Features

### 7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Acids Bases And Salts Guide
- Highlighting and Note-Taking Acids Bases And Salts Guide
- Interactive Elements Acids Bases And Salts Guide

### 8. Staying Engaged with Acids Bases And Salts Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Acids Bases And Salts Guide

### 9. Balancing eBooks and Physical Books Acids Bases And Salts Guide

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Acids Bases And Salts Guide

### 10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

### 11. Cultivating a Reading Routine Acids Bases And Salts Guide

- Setting Reading Goals Acids Bases And Salts Guide
- Carving Out Dedicated Reading Time

### 12. Sourcing Reliable Information of Acids Bases And Salts Guide

- Fact-Checking eBook Content of Acids Bases And Salts Guide
- Distinguishing Credible Sources

### 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements

---

- Interactive and Gamified eBooks

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