

## EASEL ACID AND BASE QUIZ

Multiple Choice: Select the most correct answer. (15 EK)

1. Which of the following indicates an acidic solution?

A) pH = 7.5      B) pH = 7      C) pH = 6.5  
D) more than one correct response      E) no correct response

2. Which of the following statements about weak acids is correct?

A) Weak acids always contain C atoms.  
B) A percentage dissociation of 50% is a weak acid.  
C) Weak acid molecules readily lose their acidic hydrogen atoms.  
D) more than one correct response  
E) none of the above

3. Which of the following equations represents an acid-base neutralization reaction?

A)  $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$       B)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$   
C)  $\text{Ba(OH)}_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$       D) more than one correct response  
E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

A) increases by a factor of 1.5.      B) increases by a factor of 100.  
C) decreases by a factor of 1.5.      D) decreases by a factor of 100.

5. The most basic oxides are formed from elements found in the \_\_\_\_\_ region of the periodic table.

A) right      B) left      C) center      D) top      E) none of the above

6. When  $\text{CaO(s)}$  reacts with water which of the following is true?

a) The solution contains  $\text{OH}^{-(aq)}$  and  $\text{Ca}^{2+}_{(aq)}$       b) The solution contains  $\text{O}^{2-}_{(aq)}$  and  $\text{Ca}^{2+}_{(aq)}$   
c) The solution contains  $\text{O}^{2-}_{(aq)}$ ,  $\text{OH}^{-(aq)}$ , and  $\text{Ca}^{2+}_{(aq)}$       d) The solution contains  $\text{CaO}_{(aq)}$

7. A weak base:

A) partially dissociates in water to give a little  $\text{OH}^{-(aq)}$   
B) fully reacts with water to form  $\text{OH}^{-(aq)}$   
C) completely dissociates to form  $\text{OH}^{-(aq)}$   
D) has a very high pH  
E) contains a group 1 metal and a hydroxide

8. Which of the following is NOT a property of bases?

A) They react with active metals.      B) They taste bitter and feel slippery.  
C) They have a pH greater than 7.      D) They turn red litmus blue.      E) They neutralize acids.

# Acids And Bases Test A Answers

**John T. Moore, Richard H. Langley**

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