

CHEMISTRY UNIT 1 STUDY GUIDE.

What is Matter?

- Solid, Liquid, gas
- Anything that takes up space & has mass
- Mass is the amount of matter in an object.

Types

- Substance- a particular kind of matter – pure- made up of elements and compounds
- Mixture- more than one kind of matter. 2+ elements and compounds

Properties:

Physical – property that can be observed and measure without changing the substances.

Chemical – property that can only be observed by changing the type of substance.

Extensive- only depends on the amount of matter

Mass, weight, volume, length

Intensive – only depends on the type of matter, not the amount used to identify a substance.

Color, Oder, melting/ freezing point.

Ductility- the ability to make copper into wires.

States of Matter:

	Definite volume	Definite shape	Temperature increase	compressable
SOLID	Yes	Yes	Small expans	No
LIQUID	Yes	No Takes up shape of container	Small expans	No
GAS	No	No	Large expans	Yes



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science teaching in action in Canadian schools Sites for intensive study were selected to include a diversity of both regions and school settings and at each school site various kinds of information were collected to obtain as complete a picture as

possible Major issues addressed include integration and options as forms of curriculum organization socialization as a priority the inquiry approach and understanding and the dynamics of change and dilemmas of practice *Chemistry and*

Industry Review ,1928 *Graduate Student Support and Manpower Resources in Graduate Science Education* National Science Foundation (U.S.),1969 **Annual Catalogue** University of Chicago,1919 **Independent and Weekly Review**

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