

Acids, Bases and Salts

Chemicals can be broadly classified into three categories, acids, bases and salts.

Acids

- Acids are substances which taste sour.
- They may be corrosive or non-corrosive.

Examples:

Hydrochloric acid and sulphuric acid are corrosive acids. Carbonic acid is a non-corrosive acid.

- Acids turn blue litmus red.
- There are two types of acids:
Organic acids: Naturally occurring acids obtained from plants and animals. Mineral acids: They are obtained from minerals present in the Earth's crust.

Bases

- Bases are substances which are bitter in taste and soapy to touch.

Examples: Calcium hydroxide and zinc hydroxide

- Water-soluble bases are called alkalis.
- **Example:** Sodium hydroxide
- Bases turn red litmus blue.

Indicators

An indicator is a substance which illustrates the presence of a chemical substance by changing colour. Indicators can be natural or synthetic. For example, litmus, turmeric and China rose are naturally occurring indicators, while methyl orange and phenolphthalein are synthetically produced indicators.

Natural Indicators

Litmus Paper:

Acids turn blue litmus red and bases turn red litmus blue. The solutions which do not change the colour of either red or blue litmus are known as neutral solutions.

Turmeric:

It is yellow in an acidic medium and turns reddish brown in a basic medium.

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