

6. Which of the following structures is the same as structure A?



A



A



B



C



D

7. Which of the following statements best describes the carbon-chlorine bond of the molecule below?



A. nonpolar; no dipole
B. polar; δ+ at carbon and δ- at chlorine
C. polar; δ- at carbon and δ+ at chlorine
D. ionic
E. none of the above

8. The lone pairs of electrons of the oxygen atom are located in which orbitals?



A. sp^2
B. sp^3
C. sp
D. s
E. p

9. Which statement concerning the length of bonds indicated in the molecule below is correct?



A. II is the shortest and I is the longest
B. II is the shortest and III is the longest
C. III is the shortest and II is the longest
D. III is the shortest and I is the longest
E. I is the shortest and III is the longest

10. Select the series of atoms that is correctly arranged in order of increasing electronegativity.

A. C < N < B < Br
B. P < N < As < F
C. Li < B < N < F

D. Cl < Cs < C < Co
E. Be < B < Ba < Br

11. What is the geometry of nitrogen in the following molecule?



A. tetrahedral
B. trigonal planar
C. trigonal pyramidal
D. bent
E. linear

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David R. Klein, Laurie S. Starkey

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