

**Preparing for Your ACS  
Examination  
in  
Physical Chemistry**

***The Official Guide***



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# Acs Physical Chemistry Official Study Guide

**Jianjun Gao**



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web to balance an equation that describes a reaction whether it be chemical or nuclear you must make sure that some things are conserved or remain the same on both sides of the reaction

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web balancing nuclear reaction equations exercises 1 look at the equations in the model and explain what happens to the



we can write a balanced nuclear equation for this reaction as let's do a practice example identify the product x by writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$  because the mass number is the same on both sides of the equation x must have a mass number of 0

web set up and balance a nuclear reaction equation to find the mass number and atomic number of the second nuclide  
remember that the two  ${}^{94}_{40}\text{Zr}$  neutrons each contribute

web in balancing a nuclear equation it is important to remember that the sum of all the mass numbers and atomic numbers given on the upper left and lower left side of the element symbol respectively must be equal for both sides of the equation

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$$\text{zn} + \text{s} + \text{hcl} \rightarrow \text{zncl}_2 + \text{h}_2$$

$$\text{li}_2\text{s} + \text{n}_2 \rightarrow \text{li}_3\text{n}$$

$$\text{s} + \text{ca}(\text{oh})_2 \rightarrow \text{hbr} + \text{cabr}_2$$

$$\text{h}_2 + \text{o}_2 \rightarrow \text{c}_4\text{h}_{10}$$

$$\text{o}_2 + \text{co}_2 + \text{h}_2 + \text{o} \rightarrow \text{nh}_3 + \text{cu}_2\text{h}_2 + \text{o}_2$$
 10 5 5 balance the following equations  

$$\text{fe} + \text{s} + \text{cl}_2 \rightarrow \text{fecl}_3$$

$$\text{g} + \text{c}_4\text{h}_{10} + \text{o}_2 \rightarrow \text{co}_2 + \text{h}_2 + \text{o}$$

$$\text{naoh} + \text{na}_3\text{aso}_3 + \text{h}$$

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web jan 10 2021 write a balanced equation for each of the following nuclear reactions bismuth 212 decays into polonium 212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of uranium 238 with a neutron and then spontaneously converts into plutonium 239

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web nuclear reactions are going on all around us using correctly balanced equations is important when trying to understand nuclear reactions all equations need to be balanced to conform to two conservation laws the mass number is conserved and the electrical charge of the nucleus is conserved