



Chapter 2: Acids, Bases and Salts

Science Worksheet

Standard: 10th

Subject: Science

Student Name: _____

Q1. Match the chemical substances given in Column (A) with their appropriate application given in Column (B):

Column (A)	Column (B)
A. Bleaching powder	1. Preparation of glass
B. Baking soda	2. Production of H_2 and Cl_2
C. Washing soda	3. Decolourisation
D. Sodium chloride	4. Antacid

1. A=2, B=1, C=4, D=3
2. A=3, B=2, C=4, D=1
3. A=3, B=4, C=1, D=2
4. A=2, B=4, C=1, D=3

Q2. 10ml. of a solution of NaOH is found to be completely neutralized by 8ml. of a given solution of HCl. If we take 20ml. of same solution of NaOH, the amount of HCl solution required to neutralize it will be:

1. 4ml.
2. 8ml.
3. 12ml.
4. 16ml.

Q3. Calcium phosphate is present in tooth enamel. Its nature is:

1. Basic.
2. Acidic.
3. Neutral.
4. Amphoteric.

Q4. Which of the following salts will give an aqueous solution having pH of almost 7?

1. NH_4NO_3
2. NH_4Cl
3. $CaCl_2$



Acids Bases And Salts Guide Answers

Floyd Lavern Darrow

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