



Chapter 2: Acids, Bases and Salts

Science Worksheet

Standard: 10th

Subject: Science

Student Name: _____

Q1. Match the chemical substances given in Column (A) with their appropriate application given in Column (B):

Column (A)	Column (B)
A. Bleaching powder	1. Preparation of glass
B. Baking soda	2. Production of H_2 and Cl_2
C. Washing soda	3. Decolourisation
D. Sodium chloride	4. Antacid

1. A—2, B—1, C—4, D—3
2. A—3, B—2, C—4, D—1
3. A—3, B—4, C—1, D—2
4. A—2, B—4, C—1, D—3

Q2. 10mL of a solution of NaOH is found to be completely neutralized by 8mL of a given solution of HCl. If we take 20mL of same solution of NaOH, the amount of HCl solution required to neutralize it will be:

1. 4mL
2. 8mL
3. 12mL
4. 16mL

Q3. Calcium phosphate is present in tooth enamel. Its nature is:

1. Basic.
2. Acidic.
3. Neutral.
4. Amphoteric.

Q4. Which of the following salts will give an aqueous solution having pH of almost 7?

1. NH_4NO_3
2. NH_4Cl
3. $CaCl_2$



Acids Bases And Salts Guide Answers

Floyd Lavern Darrow



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