

Name : _____ Date : _____

Acids and Bases Problems

- Without doing any calculations, identify each of the following solutions as acidic or basic:
 - $[\text{H}_3\text{O}^+] = 3.4 \times 10^{-8} \text{ M}$ _____
 - $[\text{OH}^-] = 1.8 \times 10^{-5} \text{ M}$ _____
 - $[\text{OH}^-] = 6.5 \times 10^{-12} \text{ M}$ _____
 - $[\text{H}_3\text{O}^+] = 2.6 \times 10^{-4} \text{ M}$ _____
- A solution of HBr has $[\text{H}_3\text{O}^+] = 4.5 \times 10^{-3} \text{ M}$. Calculate the pH of this solution.

- Complete the following table.

pH	$[\text{H}_3\text{O}^+]$	$[\text{OH}^-]$	Acidic, basic, or neutral?
	5.4×10^{-4}		
	1.3×10^{-5}		
10.75	1.8×10^{-11}	5.6×10^{-4}	

- If the pH of a solution is 10.3, what is the $[\text{H}^+]$ concentration?
- If the $[\text{H}^+]$ is $2.1 \times 10^{-12} \text{ M}$ HClO_4 , what is the pH? Is the solution ACIDIC, BASIC, or NEUTRAL?
- Calculate the pH if the $[\text{OH}^-]$ concentration is $5.9 \times 10^{-1} \text{ M}$? Is the solution ACIDIC, BASIC, or NEUTRAL?

Acids And Bases Practice Worksheer

Olga I. Agapova



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ChemDiscovery Teacher Edition Olga I. Agapova,2002 *Springboard: KS3 Science Teacher Handbook 2* Adam Boxer,Adam Robbins,Claudia Allan,Jovita Castelino,Thomas Millichamp,Bill Wilkinson,2024-02-09 Deliver the Springboard Science course confidently with this workload friendly approach to a knowledge rich curriculum Learn how to use cognitive science principles to deliver more effective dynamic and engaging lessons whatever your level of experience Divided into topics rather than lessons this handbook enables you to teach each topic in a responsive fashion and at a pace that is right for your students b Feel fully supported Guided explanations diagram constructions demonstrations and worked examples have been carefully crafted to support all teachers including those teaching outside of their subject specialism b Overcome common misconceptions Prerequisite knowledge checks for students help you to identify any missing knowledge or misconceptions before a topic is started with approaches to solve these covered throughout the explanations b Tailor teaching to the class in front of you Check for understanding questions allow you to adapt your delivery to meet students needs with suggested questions and responses to start the process b Take a different approach to practicals Our slow practical approach exemplifies core concepts and provides students with a clear grounding in practical skills with at least one essential practical for every unit The three Teacher Handbooks one for each year give the teacher all the guidance and detail they need to deliver great science lessons Their efficacy is further enhanced when used alongside a Boost subscription which offers supplementary guidance and materials including our customised Springboard Science two year course planner to enrich your science teaching experience Designed to be used alongside our Knowledge Book and Practice Books The Knowledge Book is the concise reference book for students covering the entire curriculum and focusing on the key facts and concepts that they need to know The three Practice books provide an abundance of questions for independent practice

Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students *Empowering Science and Mathematics for Global Competitiveness* Yuli Rahmawati,Peter Charles Taylor,2019-06-07 This conference proceedings focuses on enabling science and mathematics practitioners and citizens to respond to the pressing challenges of global competitiveness and sustainable development by transforming research and

teaching of science and mathematics The proceedings consist of 82 papers presented at the Science and Mathematics International Conference SMIC 2018 organised by the Faculty of Mathematics and Natural Sciences Universitas Negeri Jakarta Indonesia The proceedings are organised in four parts Science Science Education Mathematics and Mathematics Education The papers contribute to our understanding of important contemporary issues in science especially nanotechnology materials and environmental science science education in particular environmental sustainability STEM and STEAM education 21st century skills technology education and green chemistry and mathematics and its application in statistics computer science and mathematics education

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The School Science Review ,2005 *School Library Journal* ,1985 **Journal of the American Dietetic Association** ,1990 **Te HS&T a** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004-02 *Acid, Acid Everywhere* Center for Gifted,1996-08 Students will be able to analyze several systems during the course of the unit These include the problem system defined by the boundaries of the area affected by the acid spill the stream ecosystem into which the acid flows and the transportation system which will be disrupted by the acid spill In addition all experiments set up during the course will be treated as systems Page 3 Acids and Bases Lynnette Brent,2008-08-15 Did you know that cola is an acid And your saliva is a base Young readers will learn about common acids and bases from lemon juice to ammonia Through vivid examples and exciting illustrations this book will eagerly explore these important chemical compounds **First Year Chemistry**

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